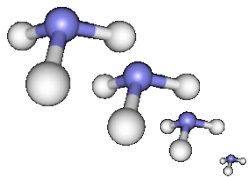
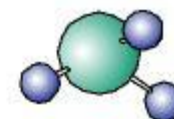


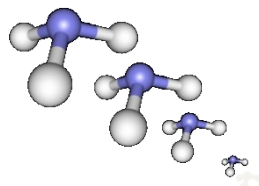
# ***Ammonia safety, a global perspective.***

***M. Kent Anderson***  
***President Emeritus, IIAR***  
***Past Chair, ASTI***



# AMMONIA - AS NATURAL AS WATER

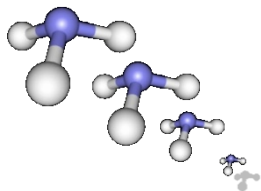




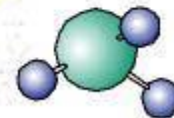
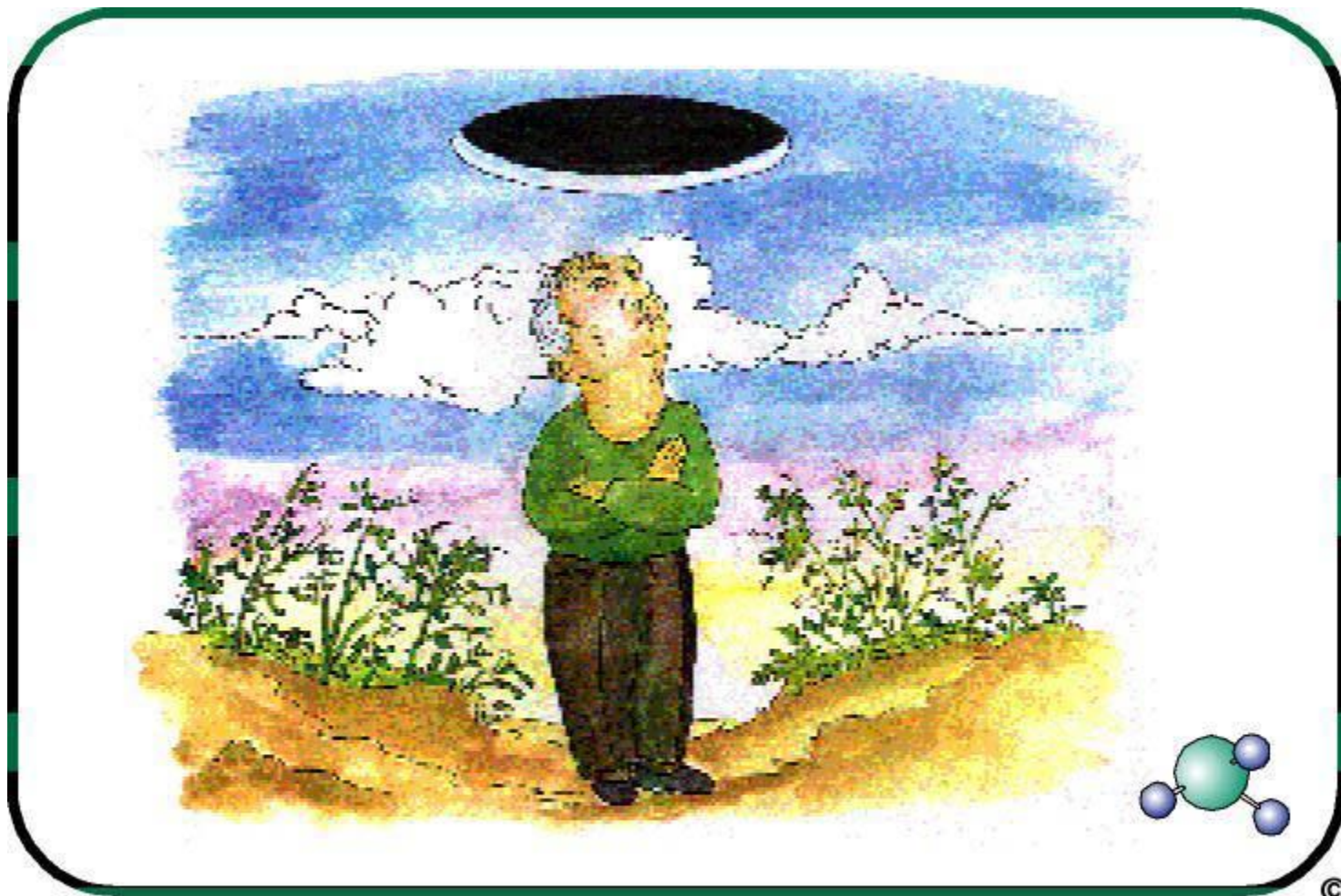
# In the beginning NH<sub>3</sub> was the catalyst for life

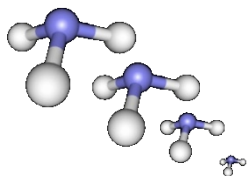


Technion Ammonia Conference. November 2017

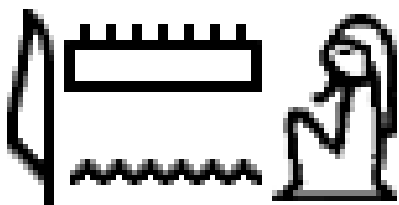
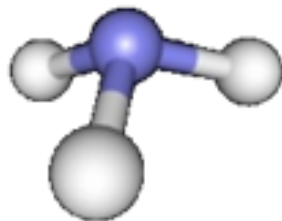


***We are in the beginning of environmental change  
Ammonia will be needed more in the future!***





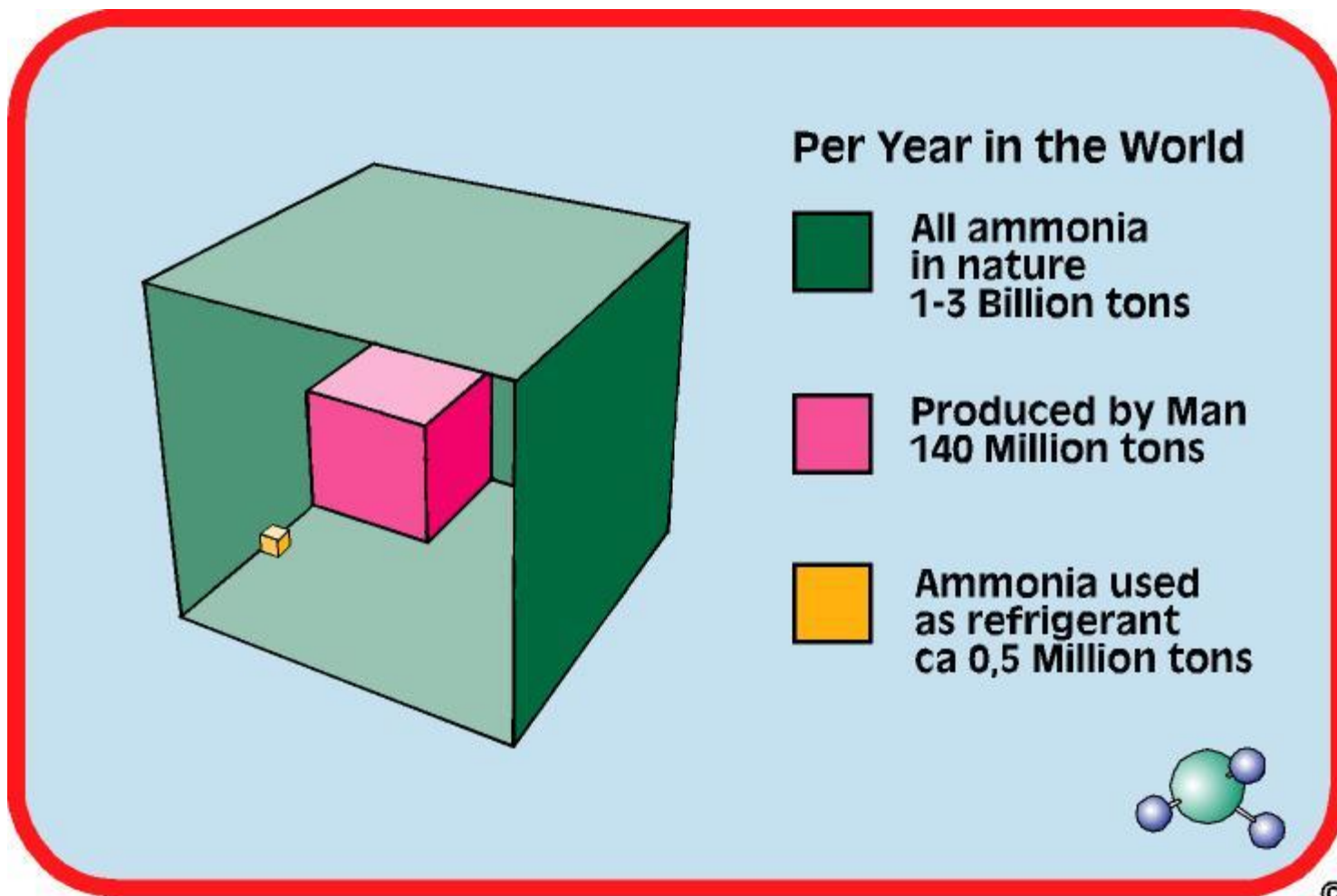
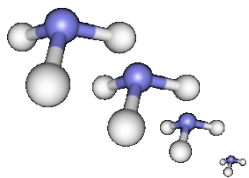
# Ammonia: What's in a Name?

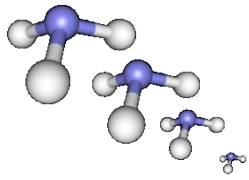


- Sal Ammoniac  
(Salt of Ammon)
- Spirit of Hartshorn
- Ammoniak/Ammoniac
- $\text{NH}_3$
- Anhydrous Ammonia
- אַמוֹנְיָה

“What’s in a name? A rose by any other name would smell as sweet”, William Shakespeare, Romeo and Juliet – Act II, Scene II

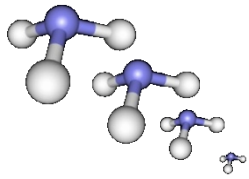






# Ammonia Production/Use

- **145 million metric tons of ammonia produced per year**
  - **China = 50 million metric tons**
  - **Russia = 12 million metric tons**
  - **India = 11 million metric tons**
  - **United States = ~10 million metric tons**
  
  - **Israel = ~120 THOUSAND metric tons (0.1% of world production capacity)**



# Ammonia Uses

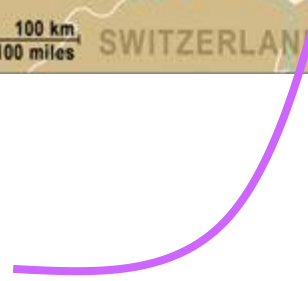
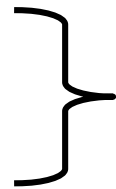
- **80 % agricultural use**
- **18 % in industrial processes**
- **<2 % as a refrigerant**





*Chilean saltpeter (potassium nitrate)*

*Peruvian guano*

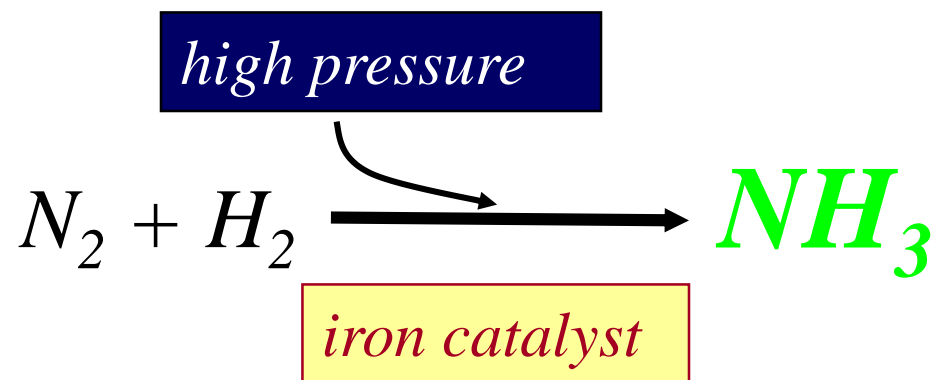
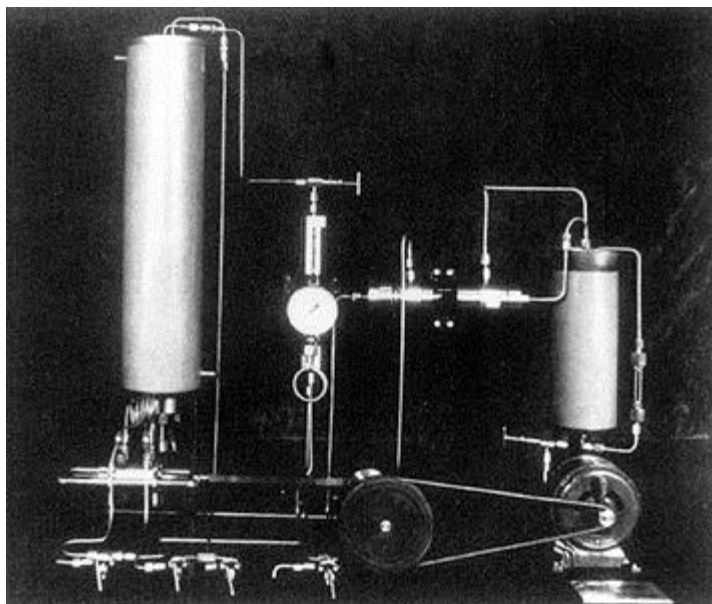


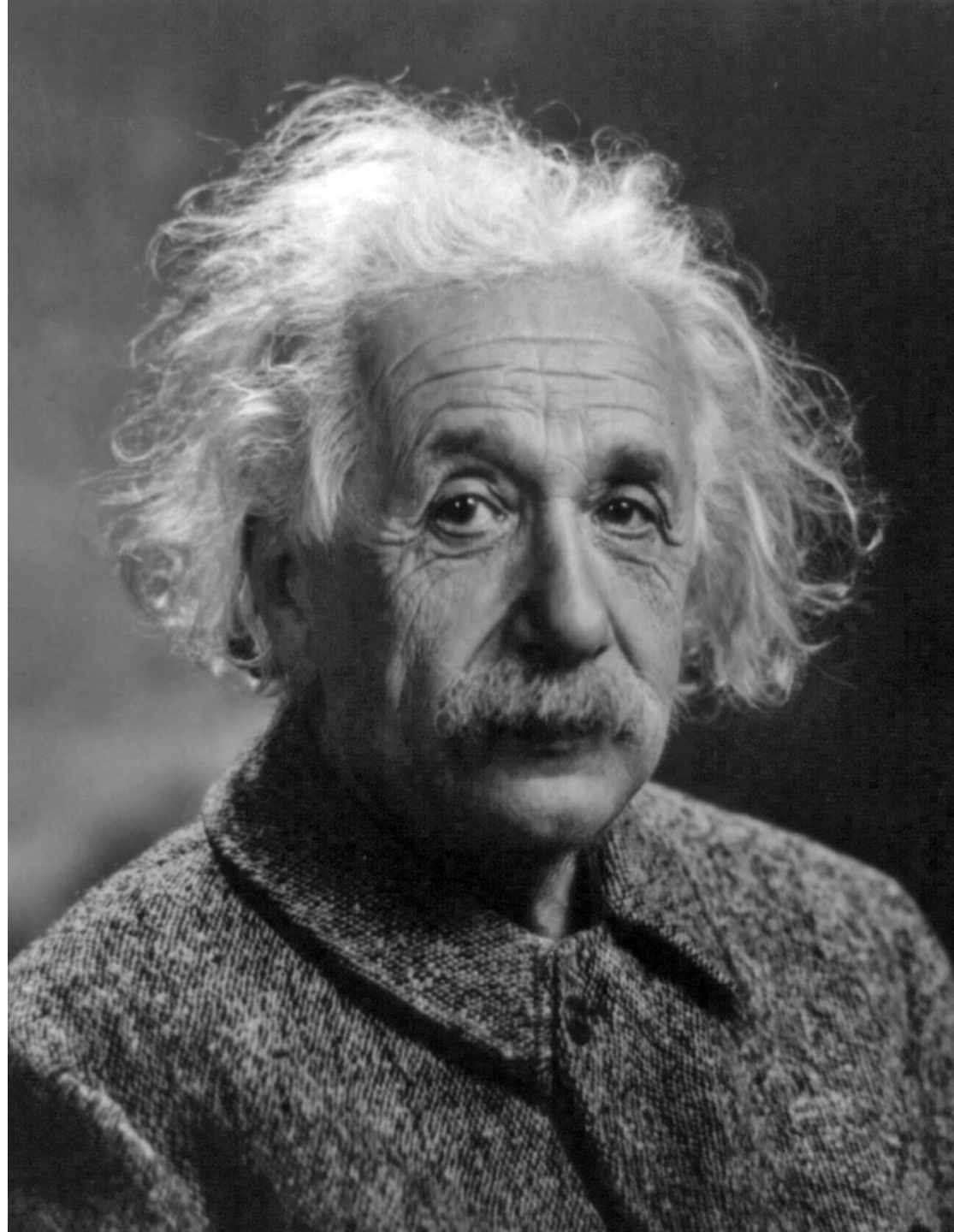


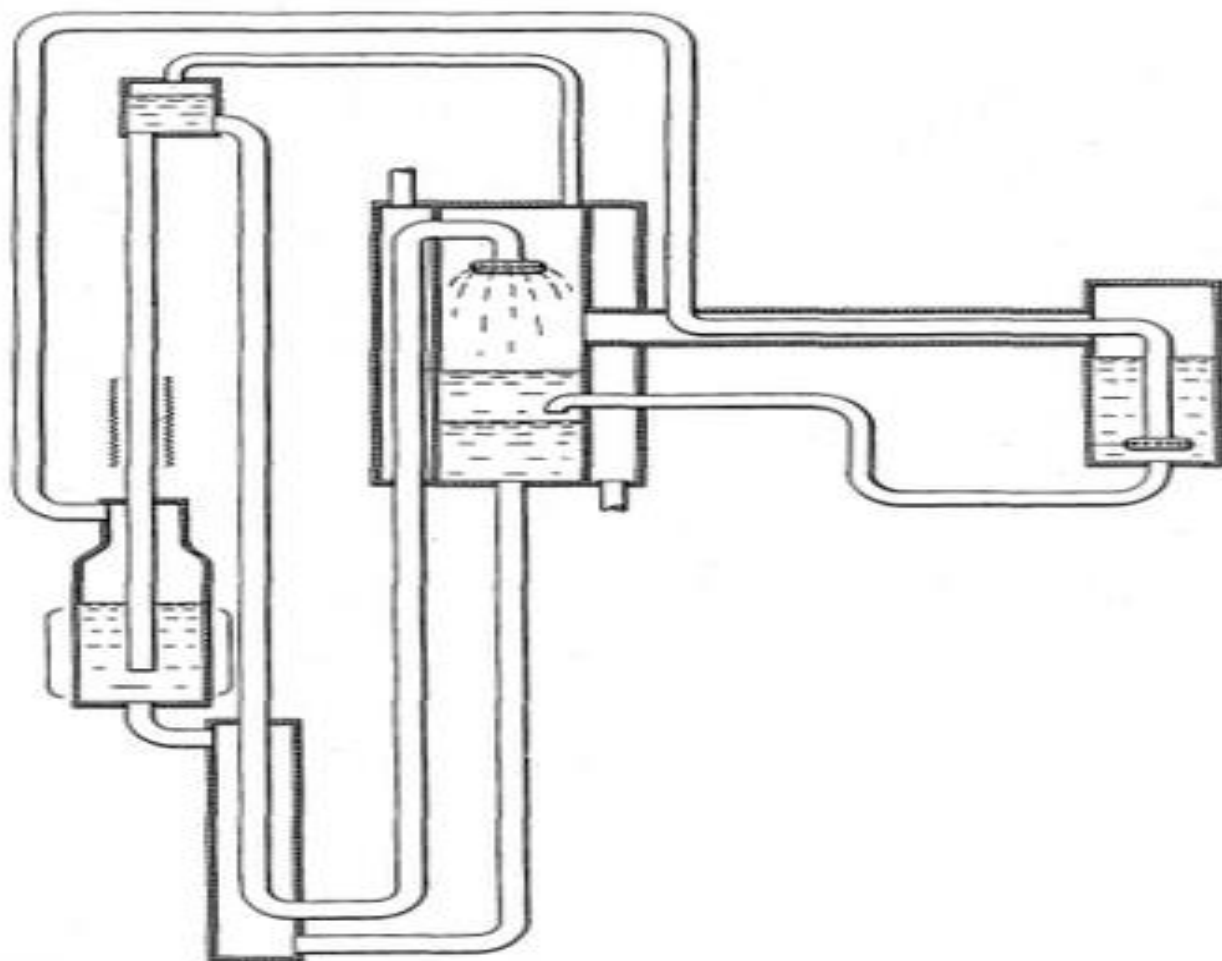
*Fritz Haber*



*Carl Bosch*



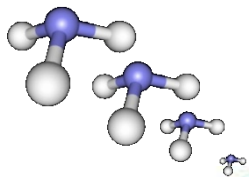




## *Einstein Refrigerator*

*Patent number US1781541 -- November 11, 1930*

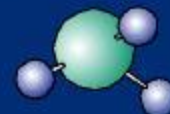
*Albert Einstein  
Leo Szilard*



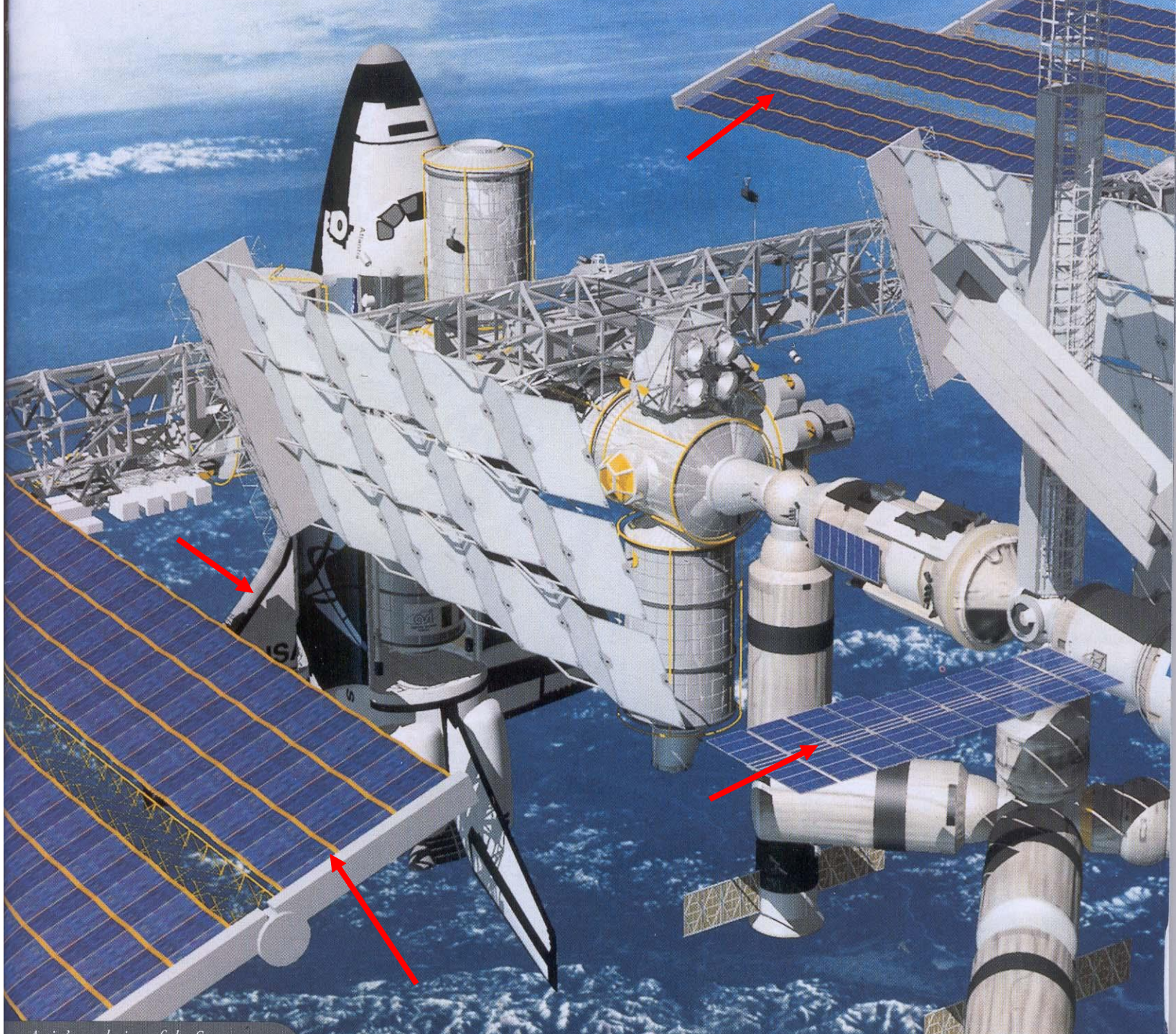
## Dr Mark Mc Linden Study for NASA 1988

860 chemicals were studied

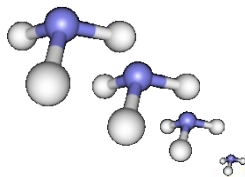
- Application
- Liquid/gas physical properties
- Thermophysical properties
- Stability
- Toxicity
- Flammability
- Compatibility to materials, gaskets, oils....
- Price





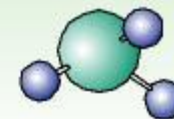


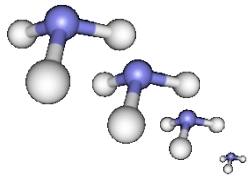




# Ammonia

- was considered the best and most appropriate followed by:
- Propane R 290, methylamine R 630, isobutane, R 152 a, R 22, Halon 1301 or R 13 B1, R 12 and R 11.
- The investigation shows that the better chemicals are simple molecules.
- It is not likely that there are other chemicals with
  - better refrigeration performance.
  - lower manufacturing costs.
  - better toxicity and flammability properties.
  - lower or no global environmental influence, ODP and GWP.





# **Ammonia Safety**

## **Hazards, Risks and Threats**

### **Hazards (Properties)**

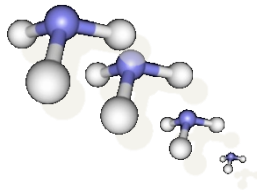
- **Chemical**
- **Physical**
- **Thermodynamic**

### **Risks**

- **Life**
- **Environment**
- **Product, Equipment, Processes**

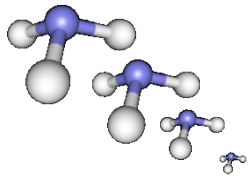
### **Threats**

- **Fire**
- **Overpressure**
- **Releases**
- **Reactivity**
- **Other (Natural, Terrorism, etc.)**



# Keys to Ammonia Safety

- Understand the Hazards
- Manage the Risks
- Control the Threats

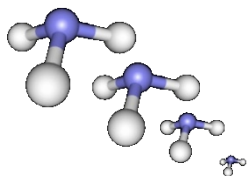


# ***Ammonia***

## ***Physical Properties<sup>1</sup>***

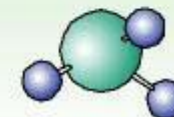
<i>Physical state</i>	<i>Colorless gas at NTP</i>
<i>Formula</i>	<i>NH<sub>3</sub>, H<sub>3</sub>N</i>
<i>Molecular Weight</i>	<i>17.03</i>
<i>Gas-density to dry air</i>	<i>0.60</i>
<i>Freezing temperature</i>	<i>-108°F (-78°C)</i>
<i>Boiling temperature</i>	<i>-28°F (-33°C)</i>
<i>Ignition temperature</i>	<i>1,204°F (651°C)</i>

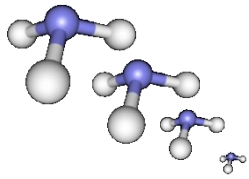
<sup>1</sup> atmospheric pressure



## THE GASES LIGHTER THAN AIR

Hydrogen	H	0,07
Helium	He	0,14
Methane	CH <sub>4</sub>	0,55
Ammonia	NH <sub>3</sub>	0,59
Hydrofluoric acid	HF	0,59
Neon	Ne	0,70
Acetylene	C <sub>2</sub> H <sub>2</sub>	0,91
Hydrocyanic acid	HCN	0,93
Carbon monoxide	CO	0,97
Nitrogen	N <sub>2</sub>	0,97
Ethylene	C <sub>2</sub> H <sub>4</sub>	0,98





# ***Ammonia***

## ***Chemical Properties<sup>1</sup>***

*Heat of combustion*      *7,992 Btu/lb (4,440 kcal/kg)*

*Heat of formation*      *-10.96 kcal/mole*

*Latent heat*              *588 Btu/lb (327 Kcal/kg)*

*Toxicity*                  *Toxic Inhalation Hazard*

*Flammable conc. in air*

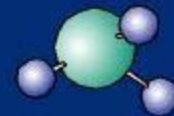
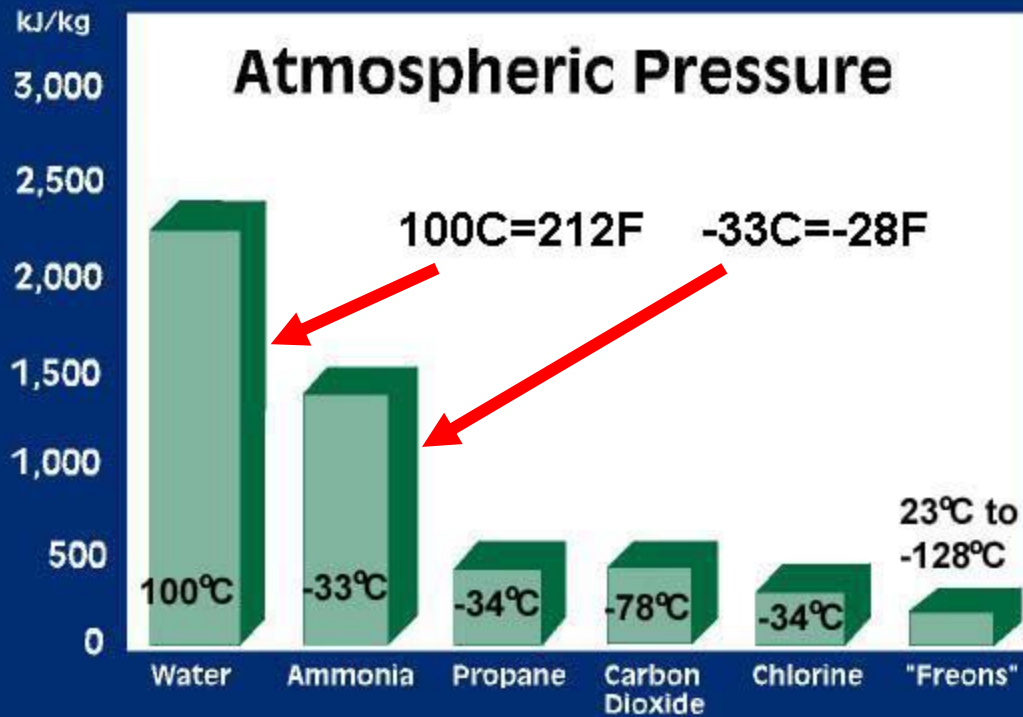
*in confined space*      *15-28 % volume*

*Ignition temperature*    *1,204°F (651°C)*

<sup>1</sup> atmospheric pressure



# Latent Heat

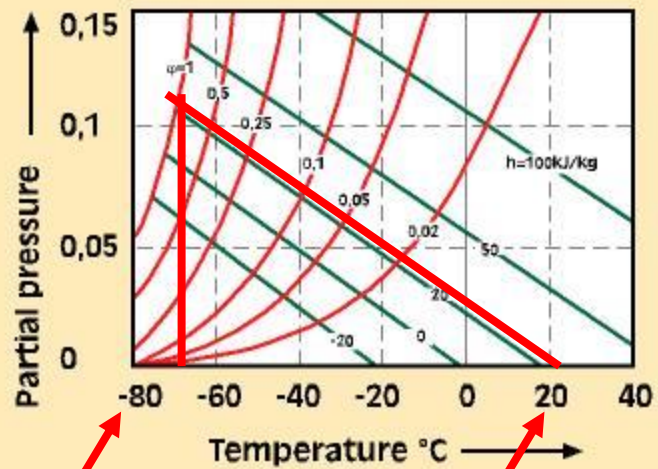
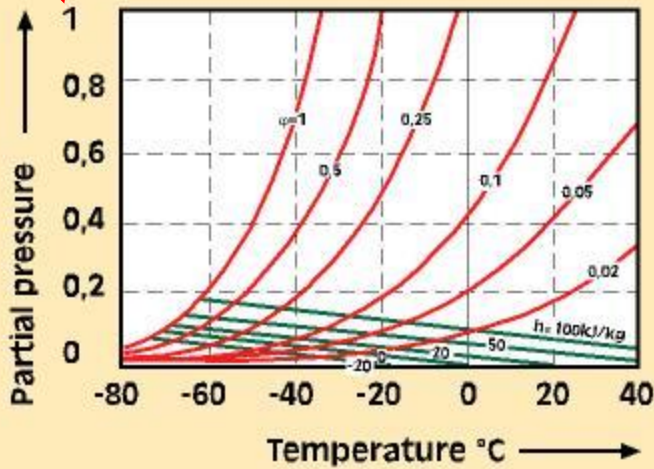


©

74

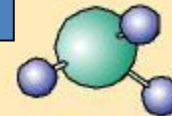
# Atmospheric Pressure

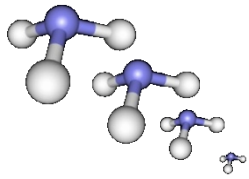
## NH<sub>3</sub> in atmospheric air



-94°F

68°F



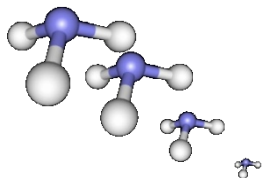


# Ammonia Toxicity & Flammability

- *PEL : 50 ppm*
- *IDLH : 300 ppm*
- *Odor threshold : 5 ppm to 50 ppm*
- *Above 100 ppm, uncomfortable*
- *Between 300 - 500 ppm, unbearable*

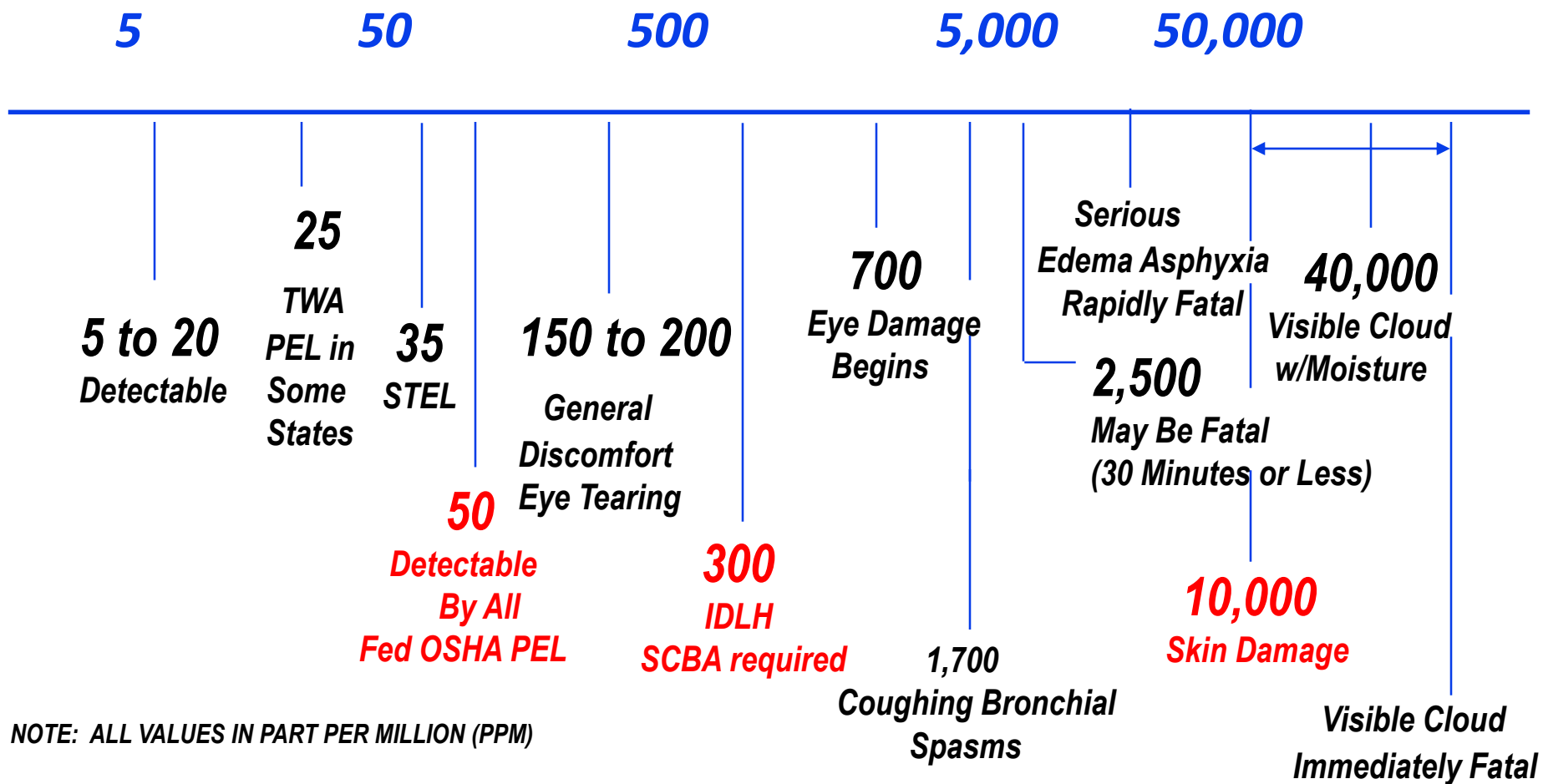
*Fire and explosion hazard : concentrations of 16% to 25%+*

*– Mixtures of ammonia and oil may reduce lower limit to 8%*



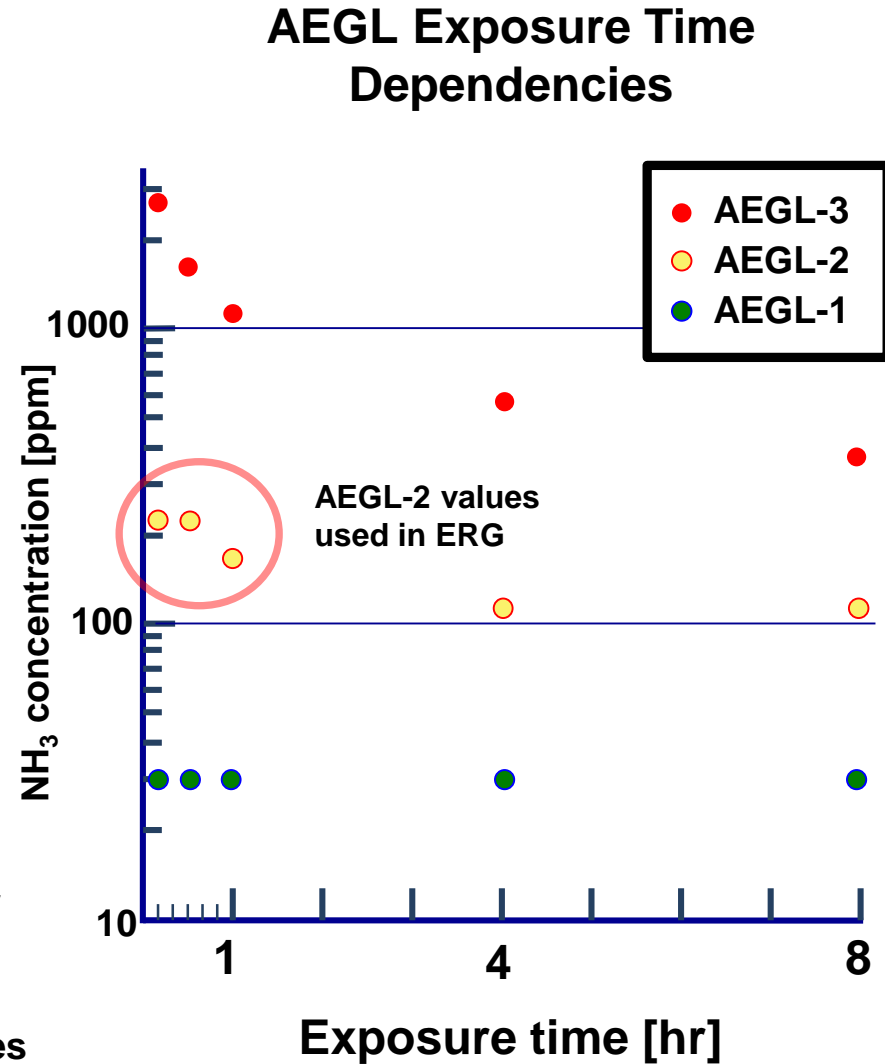
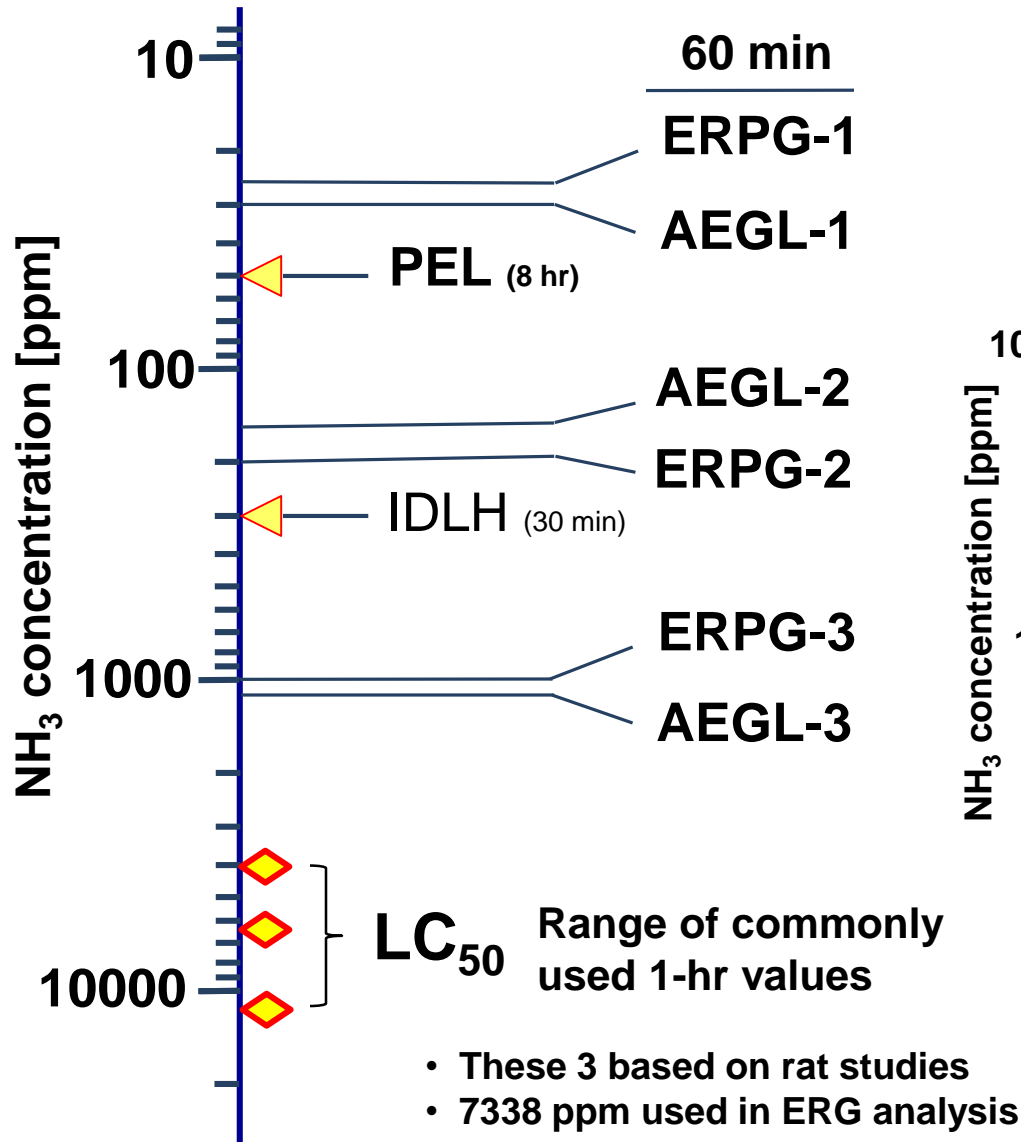
# AMMONIA **NH<sub>3</sub>** SAFETY & TRAINING INSTITUTE

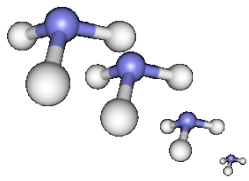
## RULE OF FIVES



NOTE: ALL VALUES IN PART PER MILLION (PPM)

# Ammonia Health Criteria





# The most basic tenant of toxicology:

*The dose makes the poison.*

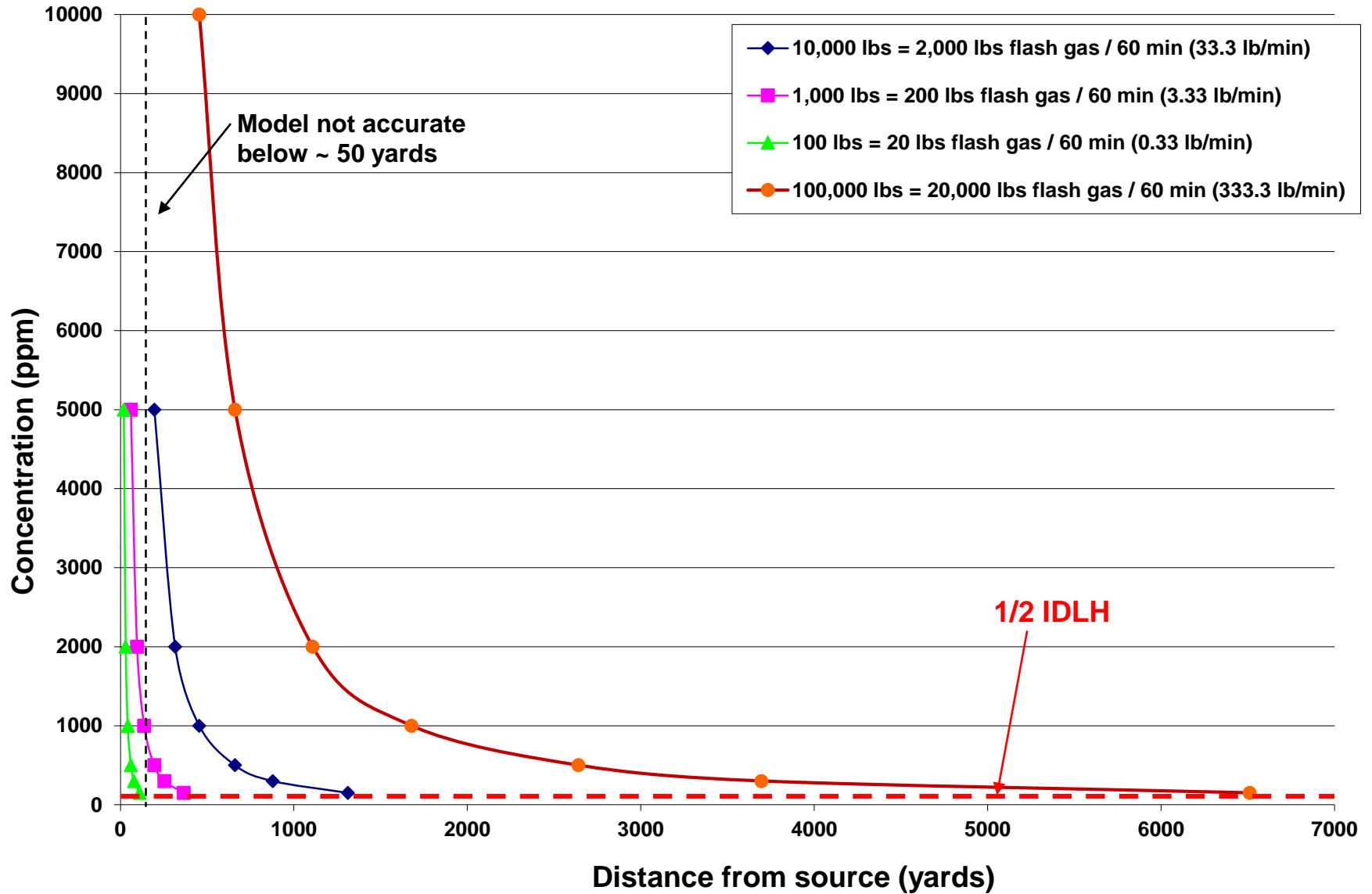
- “All substances are poisons; there is none which is not a poison. The right dose differentiates a poison and a remedy.”
- Paracelsus (1493-1541)

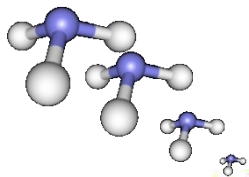




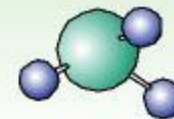
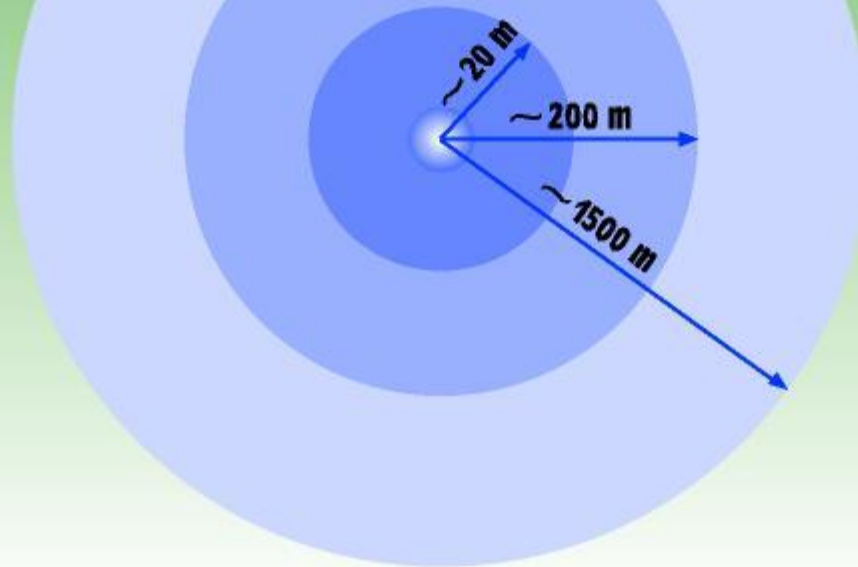
# ALOHA Concentration vs. Distance

60 min release, 20% flash, Gaussian, worst conditions



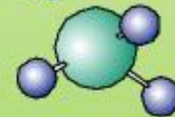
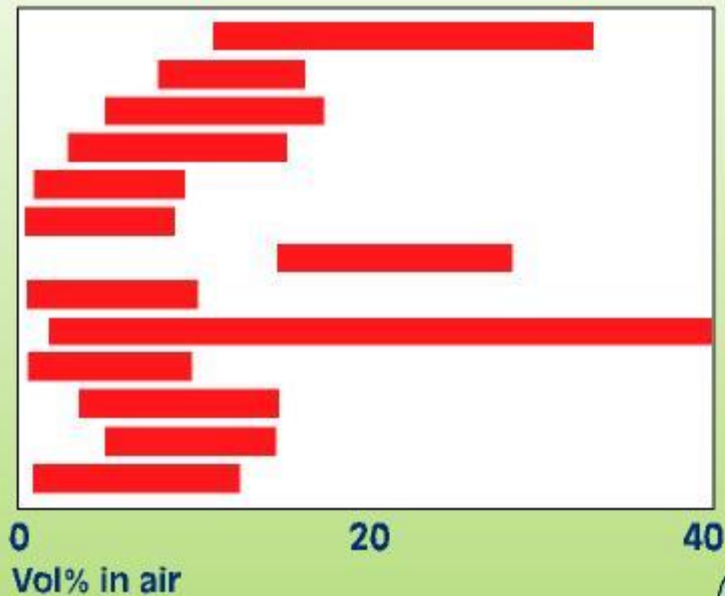


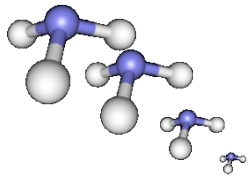
## Influence of ammonia on humans at different distances from a release.



# Comparison of different flammable concentrations in air

- R 32
- R 143a
- R 152a
- R 170 Ethane
- R 290 Propane
- R 600a Isobutane
- R 717 Ammonia
- R 1270 Propene
- Acetylene
- Petrol
- Ethanol
- R50 Methane
- Acetone

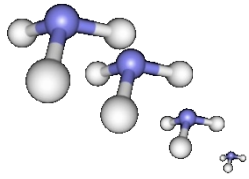




# Risks

Manage Risk to:

- Life
- Environment
- Product, Equipment and Facilities

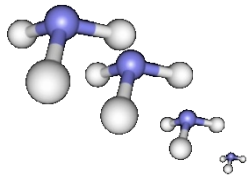


# Life Risks

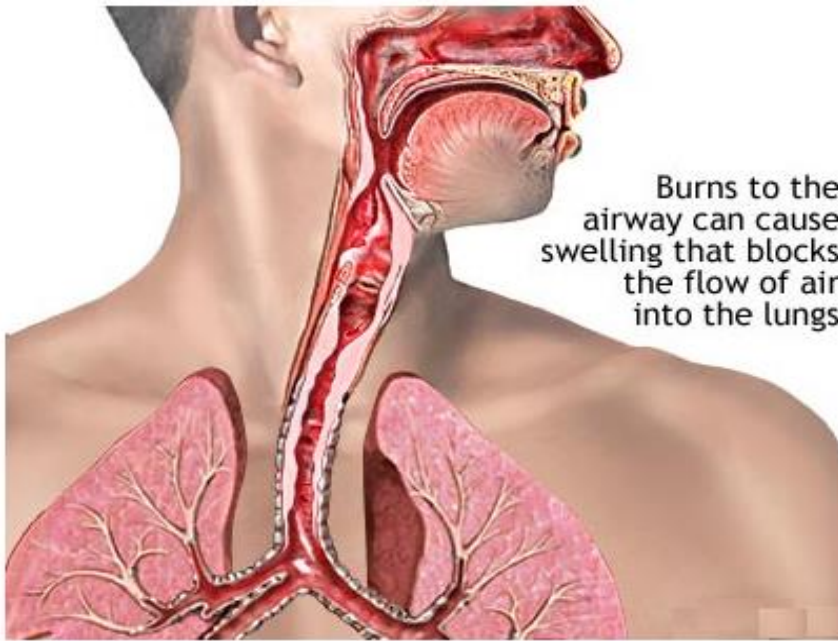
Respiratory

Exposure – skin and eyes

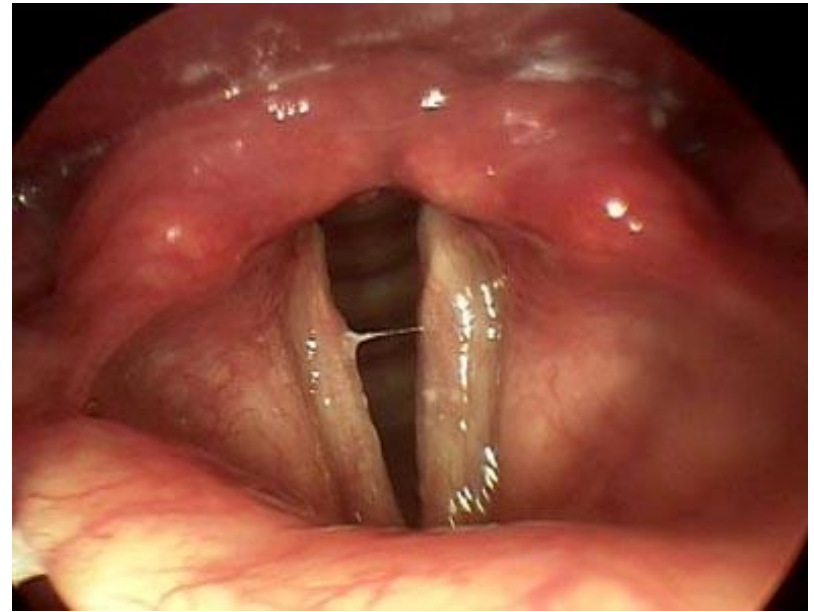
Nausea, Cardiac, Heat Stress

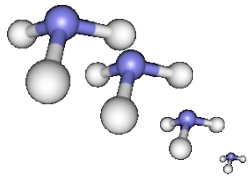


# Respiratory - Inhalation

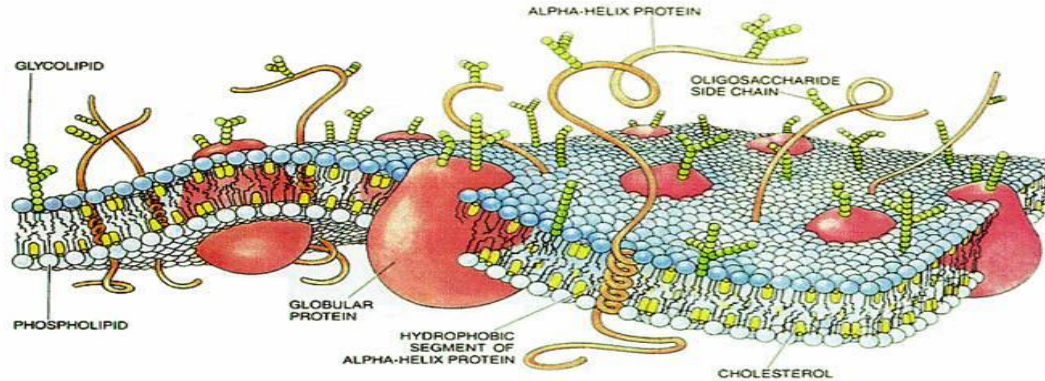


Burns to the airway can cause swelling that blocks the flow of air into the lungs



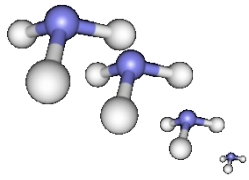


# Exposure - Skin Damage



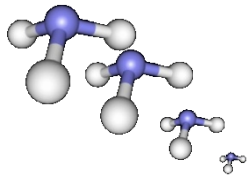
- Critical skin damage begins at 24.8°F and becomes irreversible at -18.5°F.
- The degree of tissue injury is proportional to the duration and concentration of exposure.
- Alkaline burns go deeper than acid burns.
- Alkali burns are yellow, soapy, and soft in texture. When burns are severe, skin turns black and leathery.





# Exposure - Eyes





# Medical Emergencies

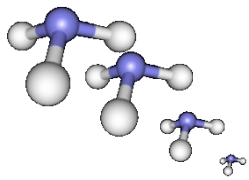


Nausea

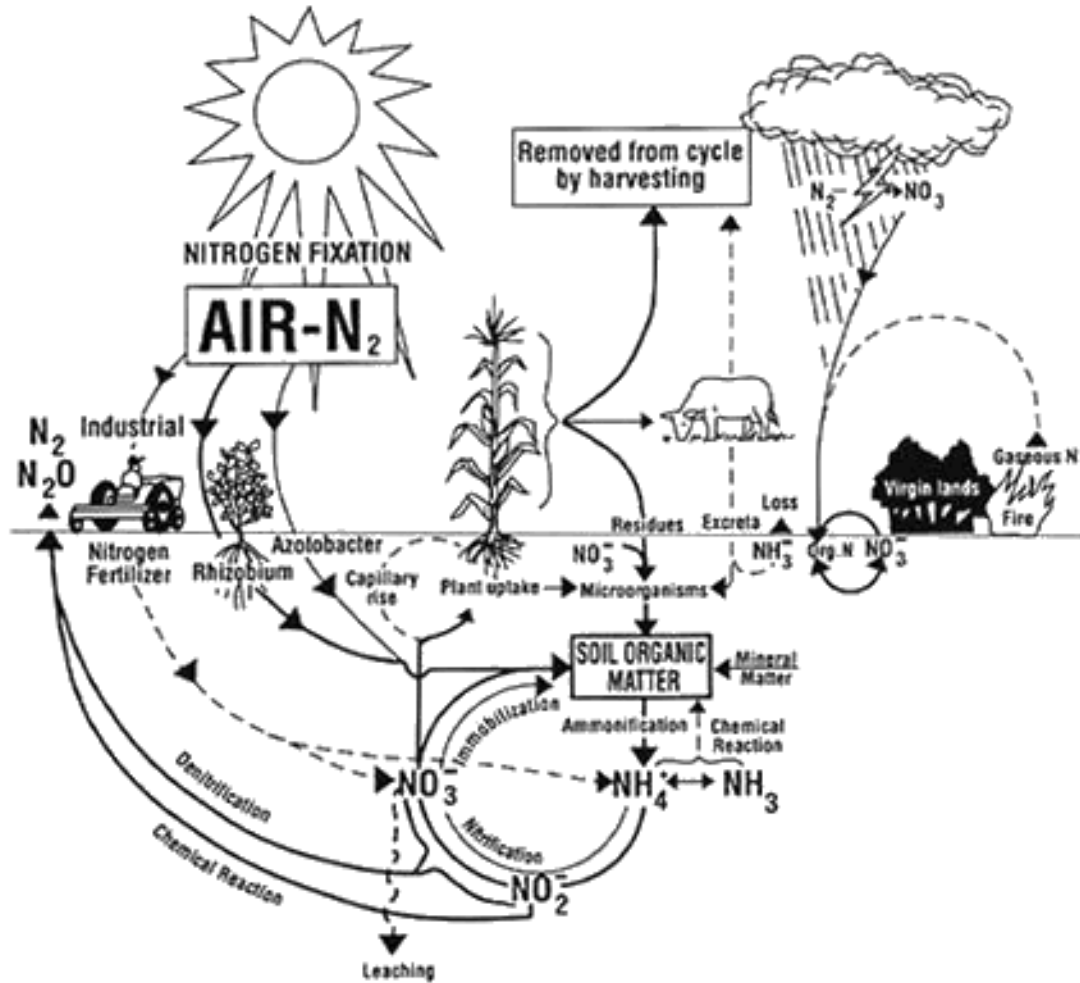
Cardiac

Heat: Rash, Exhaustion, Stroke

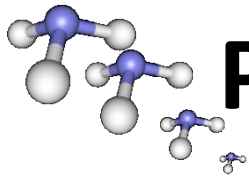
Shock



# Environmental Risks



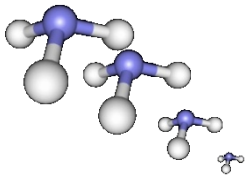




# Product, Equipment and Facility Risks

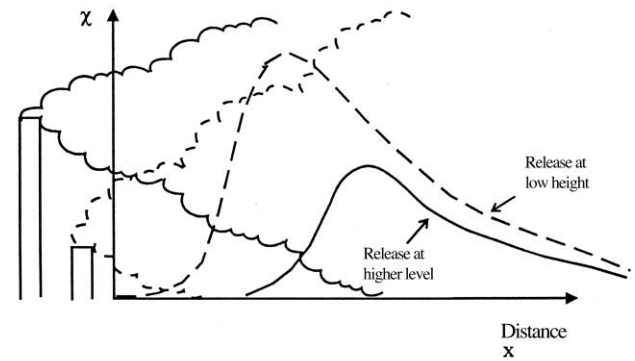
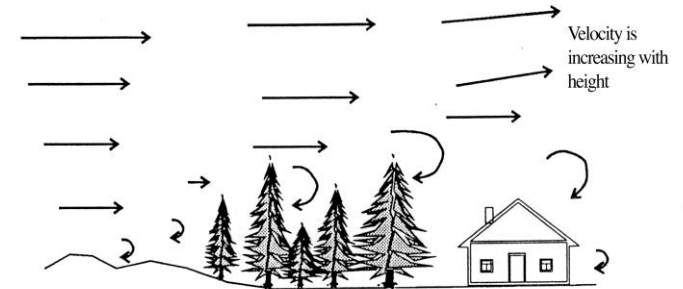
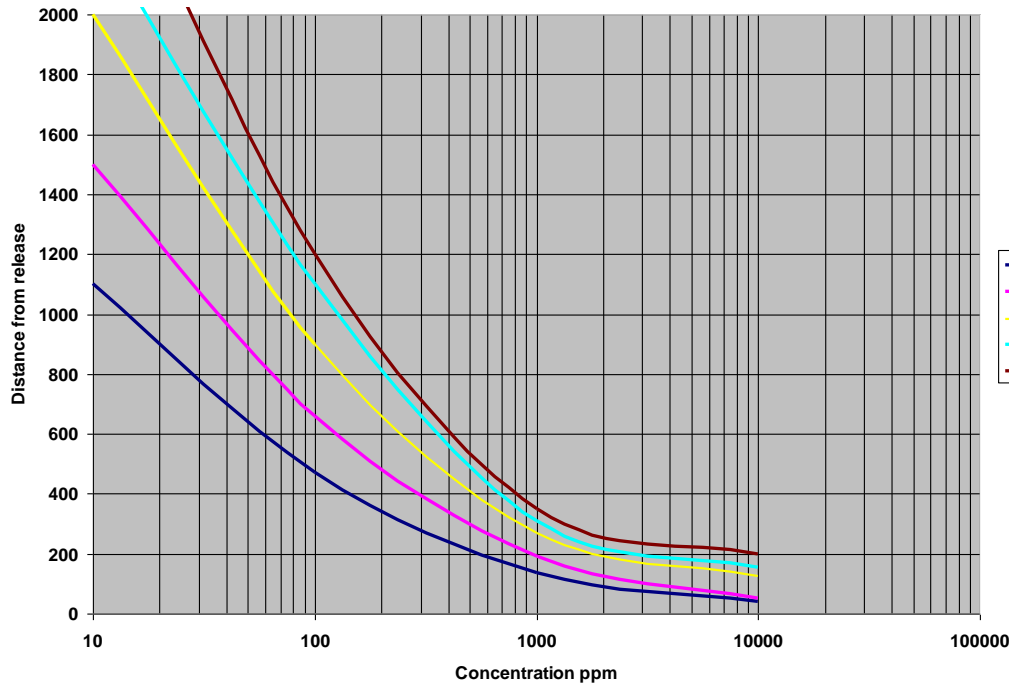


Technion Ammonia Conference. November 2017

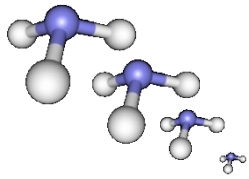


# Wind and Dispersion

Concentration in air







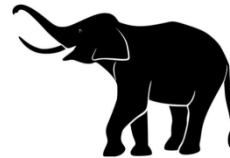
# Dealing with Risks

**There is no such thing as zero risk!**

- Black Swans<sup>1</sup>



- Black Elephants<sup>2</sup>

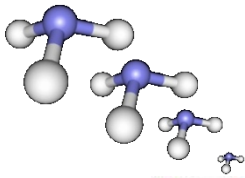


- Black Unicorns



<sup>1</sup> Aven, T. (2015). Implications of black swans to the foundations and practice of risk assessment and management. *Reliability Engineering & System Safety*, 134, 83-91.

<sup>2</sup> Möller, N., & Wikman-Svahn, P. (2011). Black elephants and black swans of nuclear safety. *Ethics, Policy & Environment*, 14(3), 273-278.



AT 35,000 FEET, THE ENGINES OF FLIGHT 430 EXPLODE FOR NO REASON!



WITH PLUMES OF DENSE SMOKE TRAILING FROM THE WINGS, THE GIANT AIRCRAFT PLUMMETS OUT OF CONTROL!



MEANWHILE, A 50-CAR FREIGHT TRAIN HITS A PENNY ON THE RAIL AT 80 MILES AN HOUR AND JUMPS THE TRACKS, DRAGGING HALF A MILLION TONS OF METAL INTO THE AIR BEHIND IT!

# calvin and hobbes

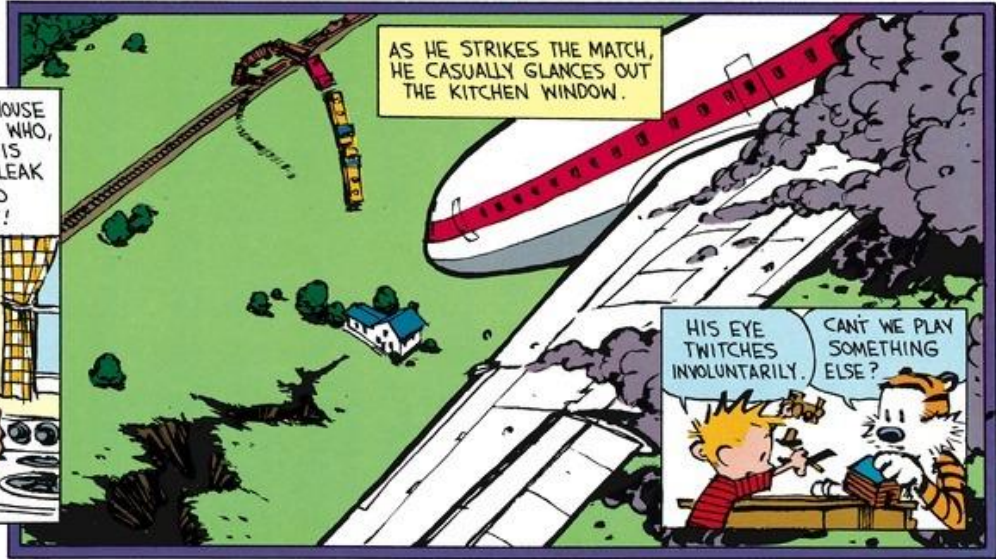
by WILSON © 1993  
distributed by UNIVERSAL PRESS SYNDICATE  
4/19



IN A FREAK COINCIDENCE, BOTH THE JET AND THE TRAIN ARE CONVERGING ON **ONE SPOT**,...WHERE TECTONIC PLATES IN THE EARTH'S CRUST HAVE JUST BEGUN TO SHIFT!



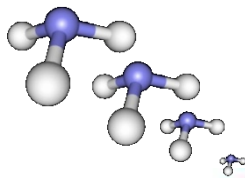
THAT SPOT IS THE HOUSE OF FARMER BROWN, WHO, AT THIS MOMENT, IS UNAWARE OF A GAS LEAK AS HE ATTEMPTS TO LIGHT HIS STOVE!



AS HE STRIKES THE MATCH, HE CASUALLY GLANCES OUT THE KITCHEN WINDOW.

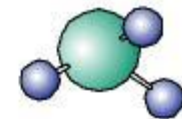


HIS EYE TWITCHES INVOLUNTARILY. CAN'T WE PLAY SOMETHING ELSE?

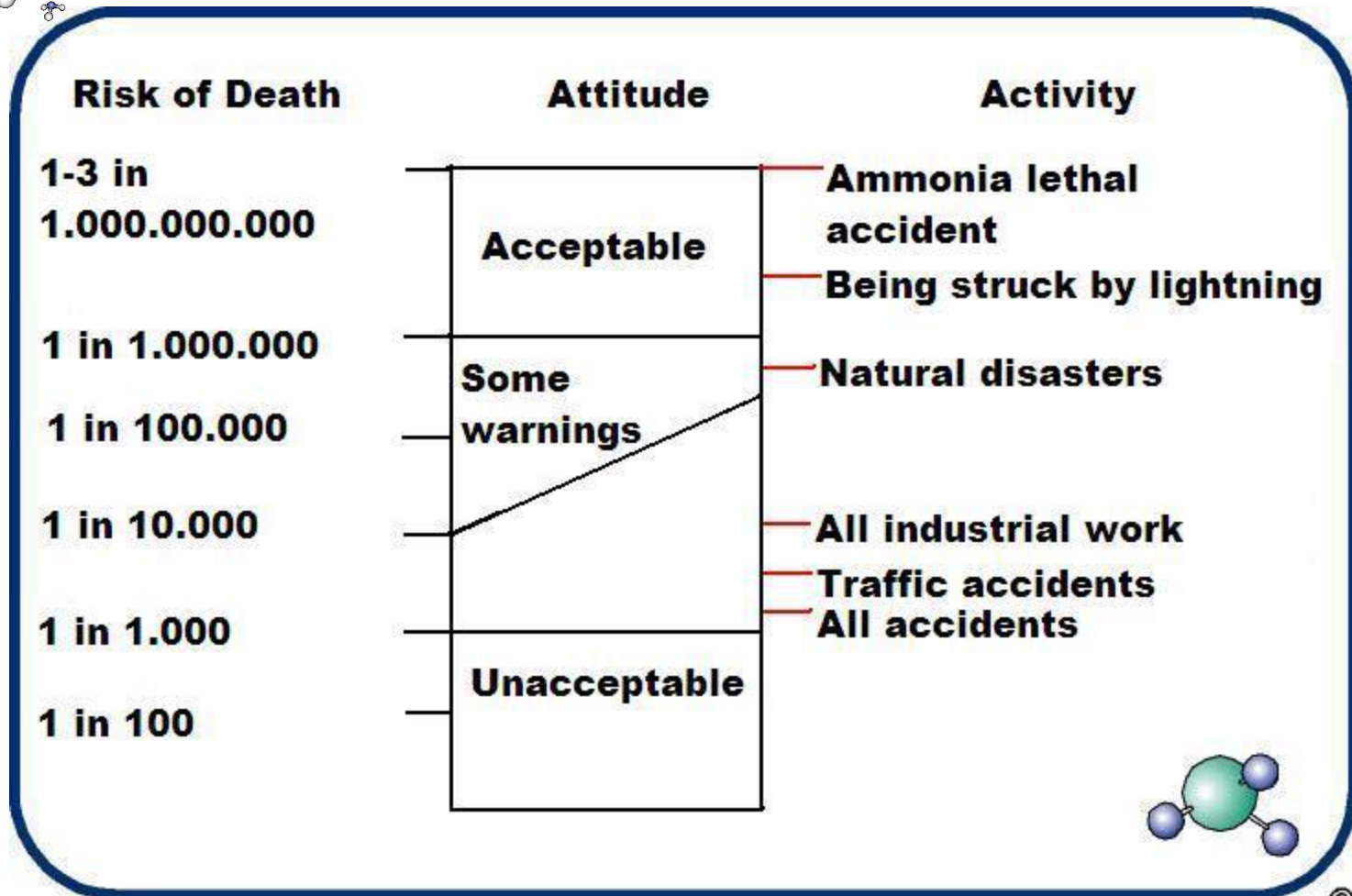
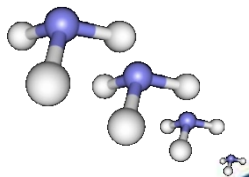


# RISK EVALUATION

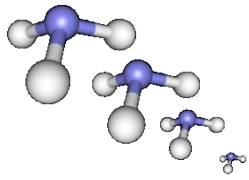
Occasion per plant and year		CONSEQUENCE $\longrightarrow$				
		Near-miss	Incident	Accident	Catastrophe	
<b>FREQUENCY</b>	20 Common	A	A	A	A	<div style="display: flex; flex-direction: column; align-items: flex-start;"> <div style="display: flex; align-items: center; margin-bottom: 5px;"> <span style="background-color: #FFD700; padding: 2px 5px; margin-right: 5px;"><b>A</b></span> <p>Comprehensive Planning (CP) mandatory</p> </div> <div style="display: flex; align-items: center; margin-bottom: 5px;"> <span style="background-color: #FF00FF; padding: 2px 5px; margin-right: 5px;"><b>B</b></span> <p>CP optional</p> </div> <div style="display: flex; align-items: center;"> <span style="background-color: #008080; padding: 2px 5px; margin-right: 5px;"><b>C</b></span> <p>CP may be unwarranted and unnecessary</p> </div> </div>
	1 Likely	A	A	A	A	
	$10^{-2}$ Reasonably likely	C	B	B	A	
	$10^{-4}$ Unlikely	C	B	B	A	
	$10^{-6}$ Very unlikely	C	C	C	B	







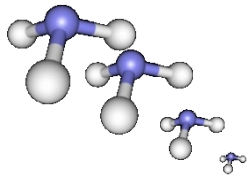
©



# ***Available facts about fatal accidents related to ammonia***

- ***Number of fatalities is less than 2 persons per 1,000,000,000 (billion) population.***
- ***Such low figures do not normally justify extraordinary regulatory measures.***
- ***Victims are those closest to a release, exposed to very high concentrations, and unable to escape or shelter-in-place.***
- ***Most fatalities are to workers, not responders or the public and are preventable.***

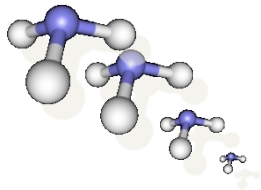




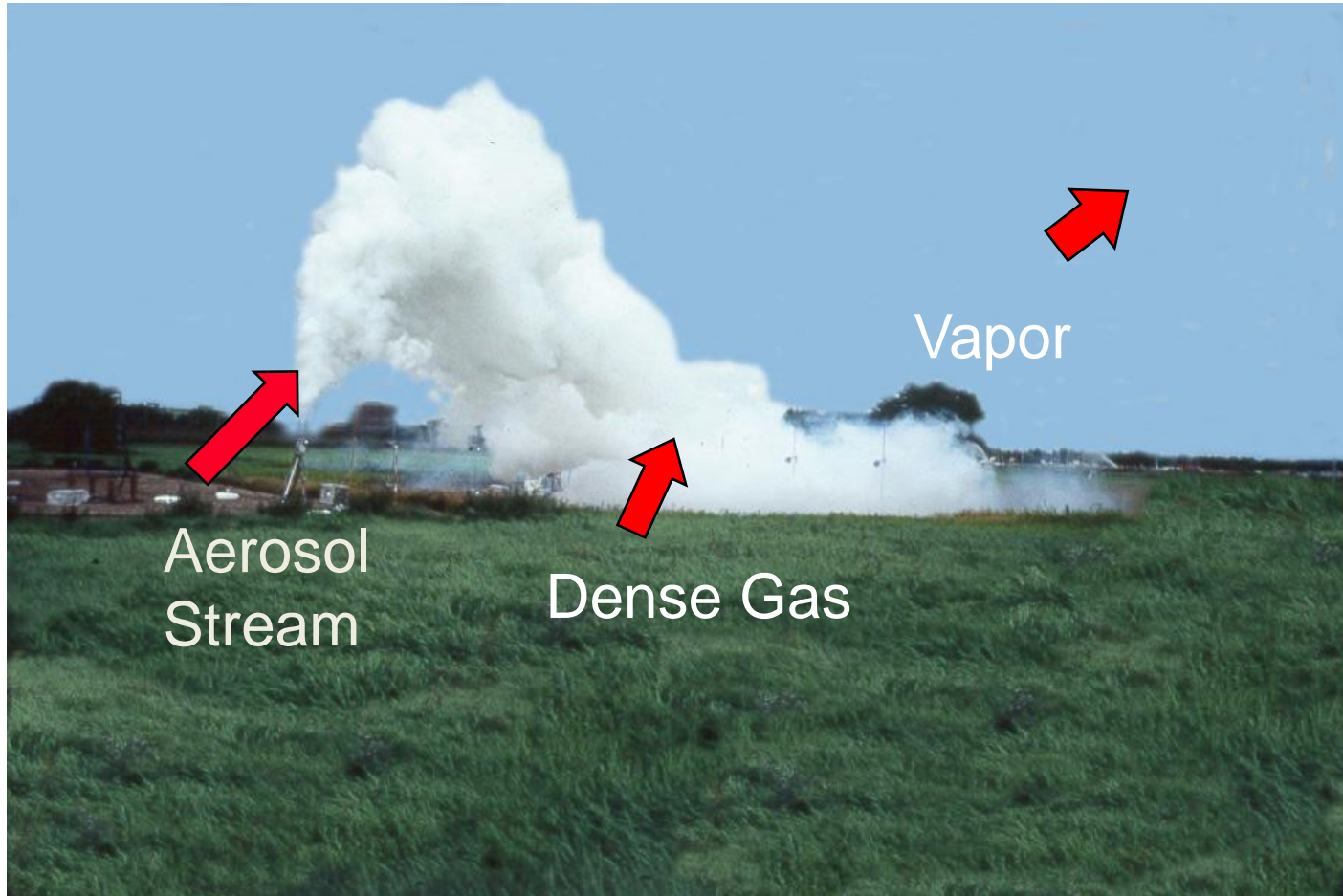
# Threats

Control the Threats from:

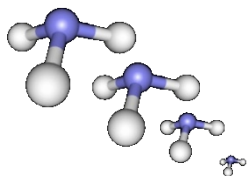
- Fire
- Overpressure
- Release
- Reactivity
- Other (Natural, Terrorism, etc.)



# Types of Release

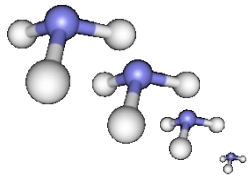


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# *Chemical Regulatory History*

- Major Accidents
  - Flixboro (June 1974)
  - Seveso (July 1976)
  - Love Canal (1978)
  - Times Beach (1982)
  - Bhopal (December 1984)
  - West, TX (April 2014)
- Regulations
  - Occupational Safety
  - Environmental
  - Transportation
  - Chemical Safety/Risk Management
  - Homeland Security
- Codes and Standards



# Terra Nitrogen

Sioux City, IA

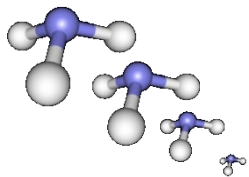
- December 1994
- AN process explosion
- 5,700 Ton release
- 4 dead (not from NH<sub>3</sub>)



Photo 8 - 12  
Water Tank Damage (Ammonia Tank Background)



Photo 8 - 13  
Ammonia Bullet Tank Damage

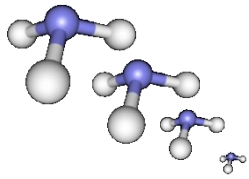


# Minot

Minot, ND

- January 2002
- Rail car derailment
- 5 tank cars ruptured
- 146,700 gal release
- 1 dead, 11 injuries
- 1 car travelled 1,200'
- +74,000 gal released over the next 6 days

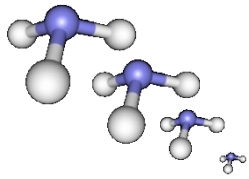




# Why You Shouldn't Believe Everything You Read On The Internet

Something important always gets left out.





# ***“Explosion Highlights Dangers of Anhydrous Ammonia”***

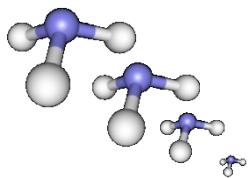
The fire at a fertilizer plant in West, Texas killed between 5 and 15 people.

By Ker Than, for [National Geographic News](#)

PUBLISHED April 20, 2013

**A fertilizer plant in the community of West, Texas that exploded on Wednesday to deadly effect was known to produce and store a volatile and potentially dangerous form of nitrogen-based fertilizer known as anhydrous ammonia.**

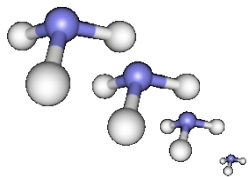
Wednesday's explosion, which happened around 7:50 p.m., suggested otherwise. Beyond the death toll, which authorities said was unclear, **the blast injured dozens of people and was compared to a "nuclear bomb" by some witnesses.**



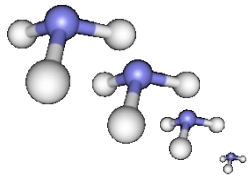
West Tx Fertilizer Plant  
Explosion  
April 17, 2013, 1953 hours



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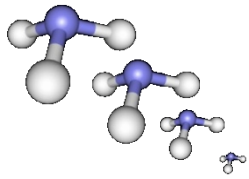


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# **LESSONS LEARNED FROM AMMONIA RELEASES**

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# ***Ammonia accidents and incidents***

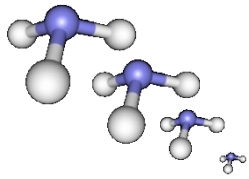
***Ammonia releases are common.***

***Ammonia accidents that cause significant consequences (fatalities, injuries, property damage, off-site consequences, environmental damage, etc.), however, are rare.***

***There are common/typical sources and causes of ammonia incidents.***

***Nearly all ammonia incidents can be prevented.***





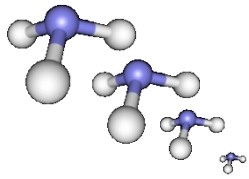
# *Accident /Safety Triangle\**



\* *H. W. Heinrich, "Industrial Accident Prevention", 1931*

*Frank E. Bird, Jr., "Practical Loss Control Leadership", 1985*

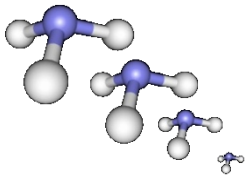
Technion Ammonia Conference. November 2017



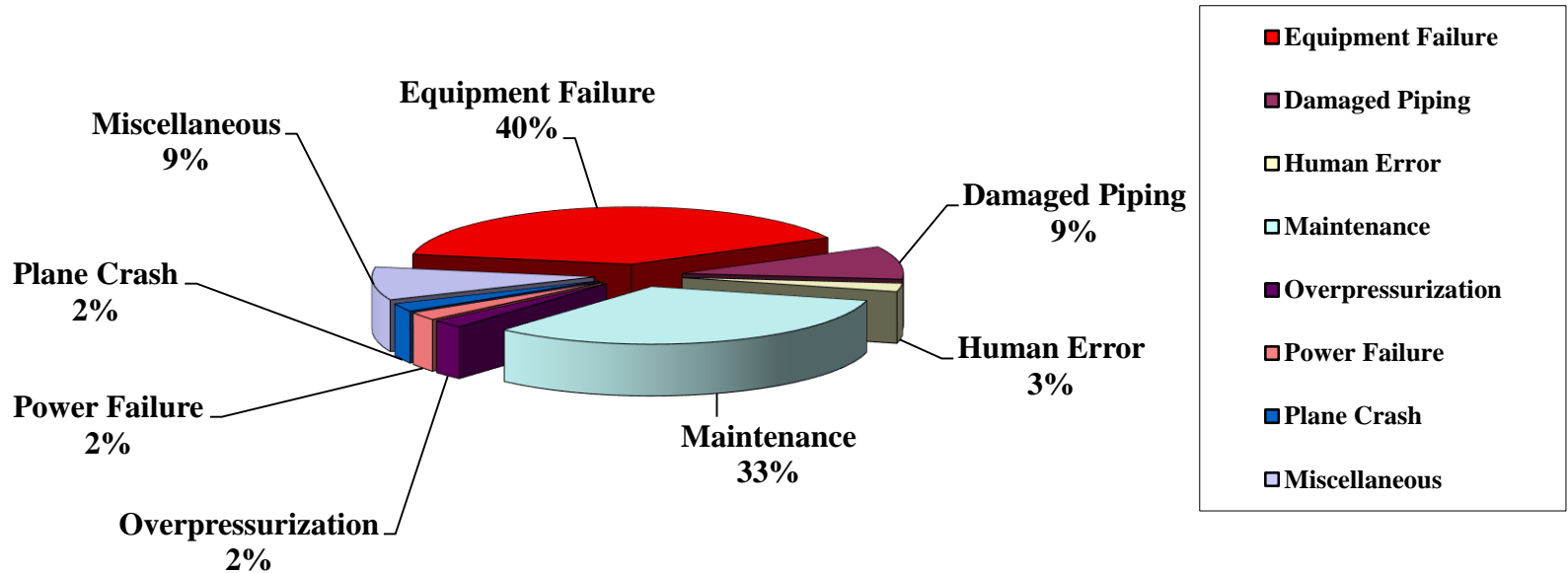
# **NH<sub>3</sub> Accident Research**

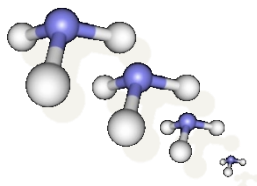
## **Collaboration with US Chemical Safety Board**

- **HMIS database (US OSHA)**
- **RMP database (US EPA)**
- **Regulatory reporting (HSEES - CDC/HMIS -DOT)**
- **State regulatory search**
- **International**
  - **MARS (EU – Seveso Directive)**
  - **ARIA/BARPI (France)**
  - **HSE Accident Data (United Kingdom)**

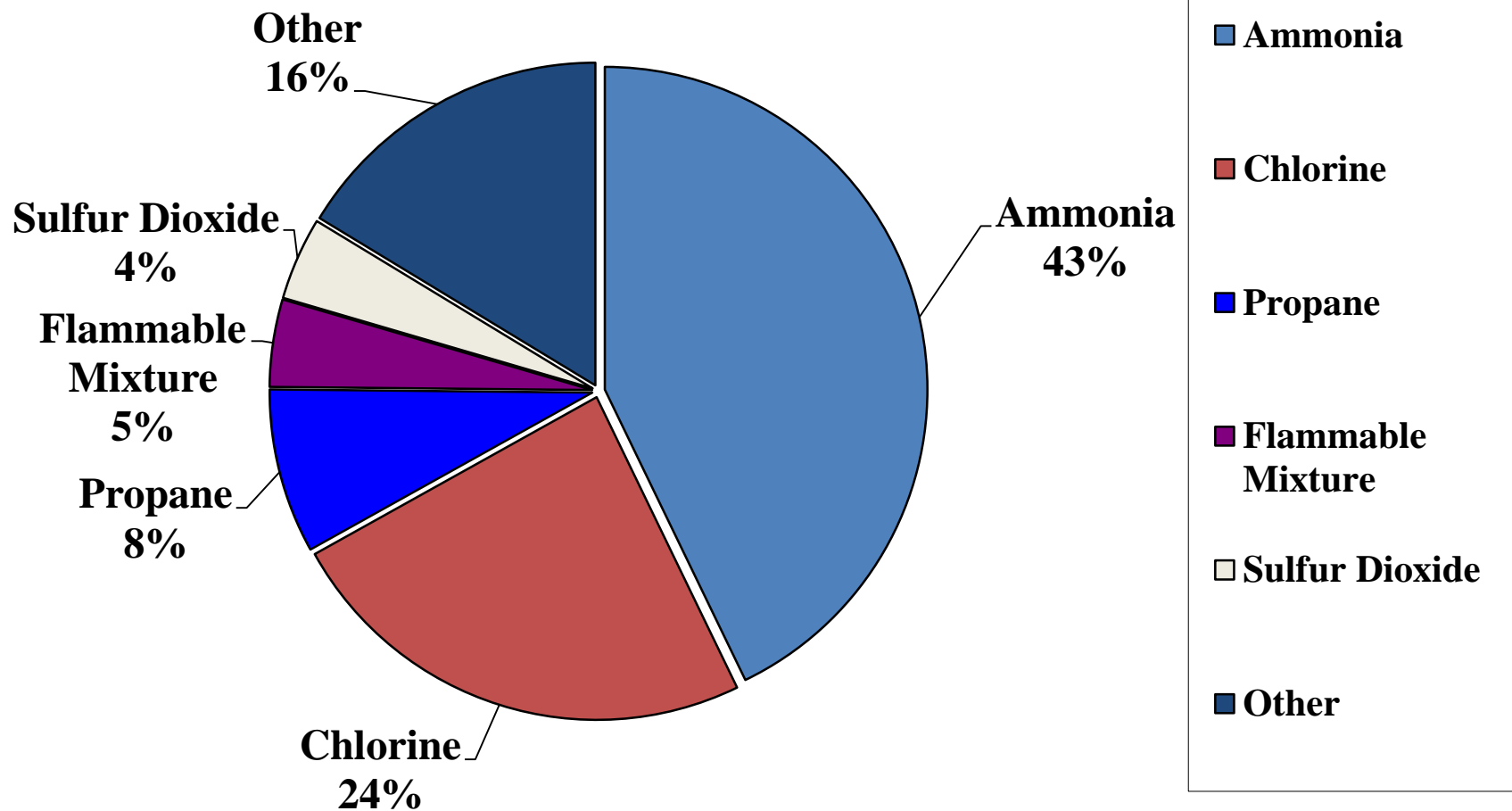


# Ammonia Accident Causes - OSHA

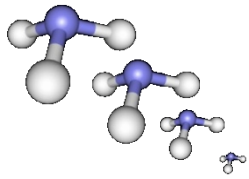




# ***RMP Database Top 5 Chemicals***



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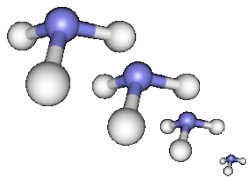
# **RMP Database 1994-2013**

- **7,619 RMP covered facilities with reportable ammonia inventory**
- **Over 2,000 incidents involving ammonia (57%)**
- **Over 1,200 Ammonia refrigeration incidents**

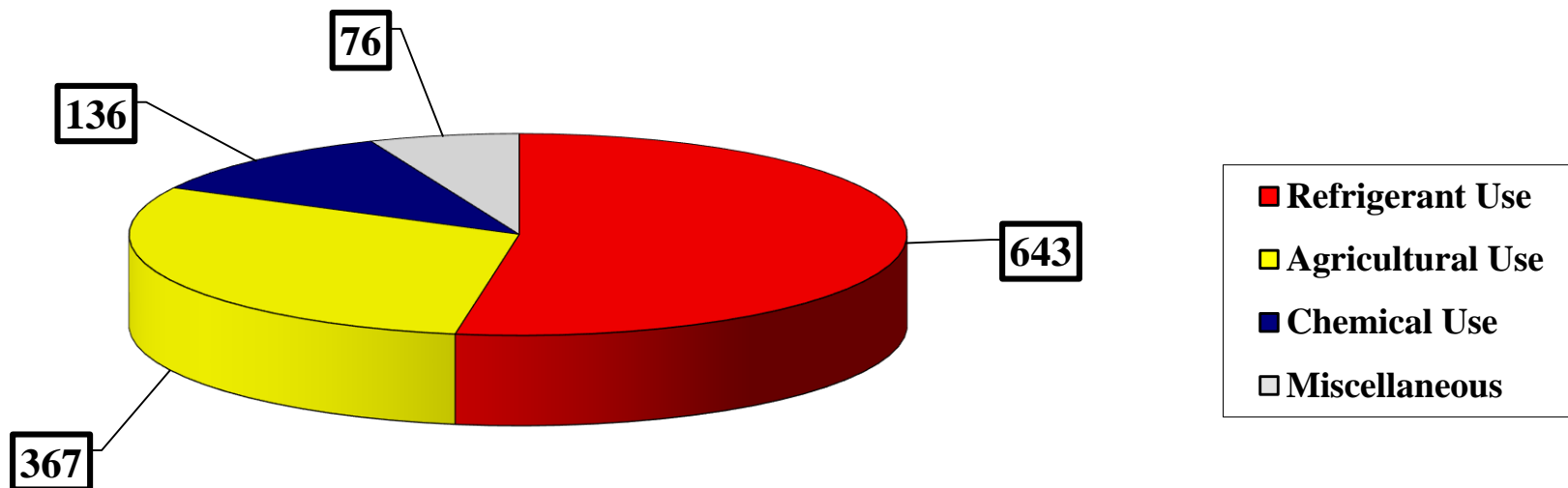
## **Ammonia is THE #1**

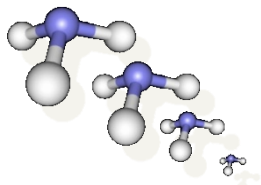
- ✓ **RMP/PSM/CFATS Covered Chemical**
- ✓ **Chemical for RMP 5-year Accident History**



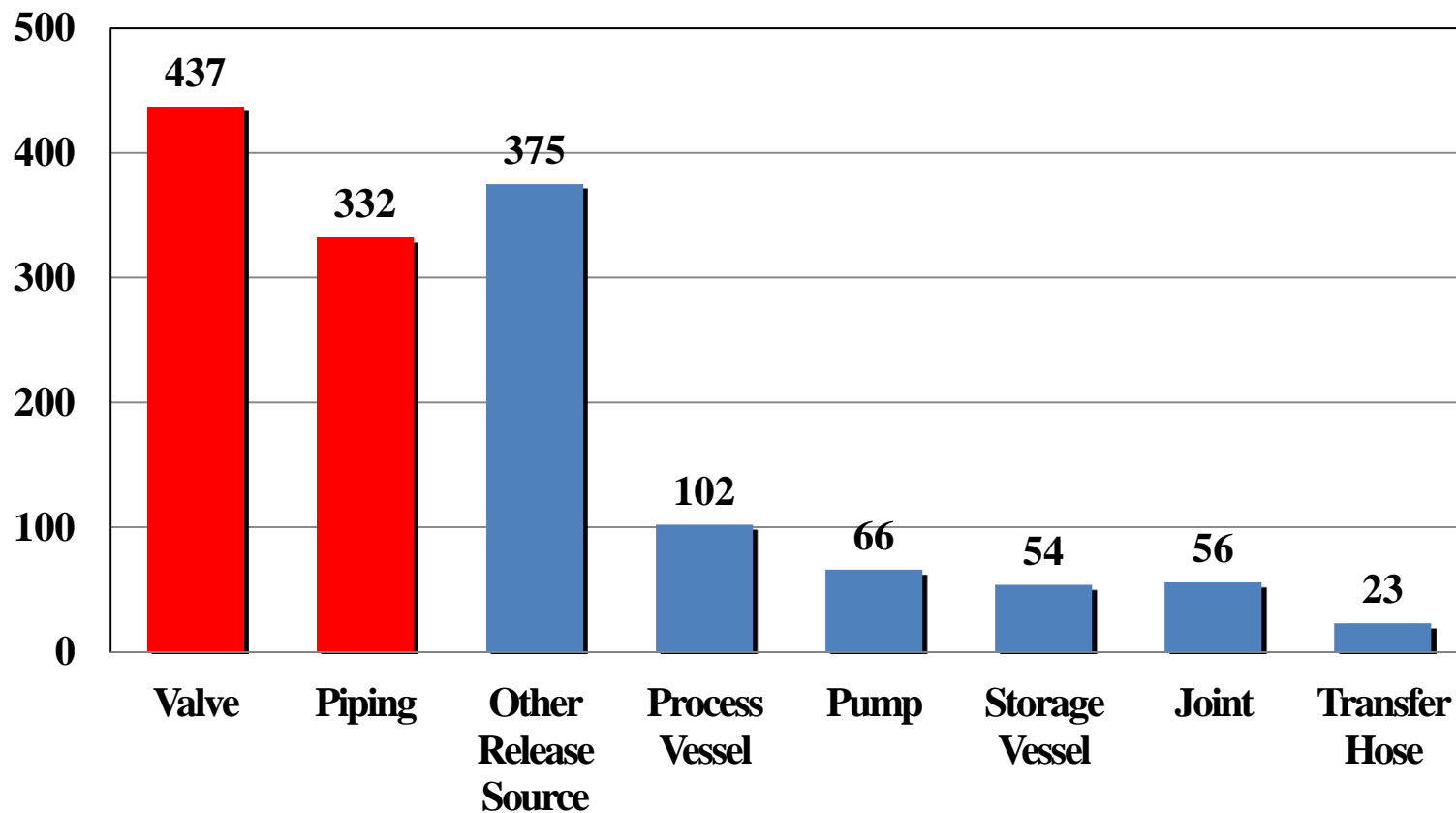


# RMP Ammonia Incidents by Ammonia End Use 1994-2004

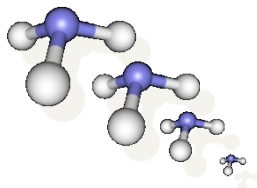




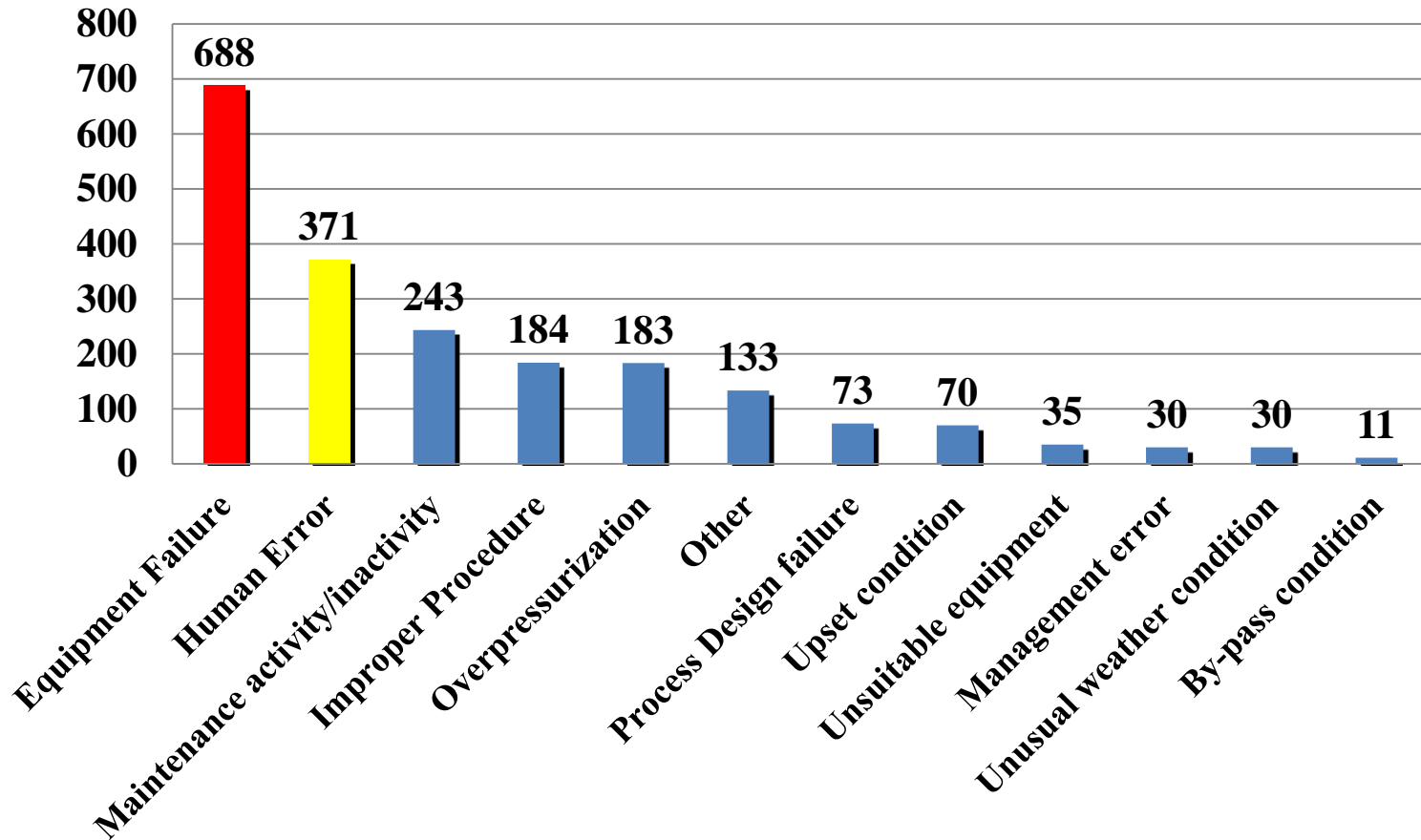
# RMP Ammonia Incidents Release Sources (1994-2013)



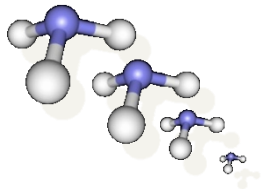
1,253 Total Releases



# RMP Ammonia Incidents Reasons for Release (1994-2013)

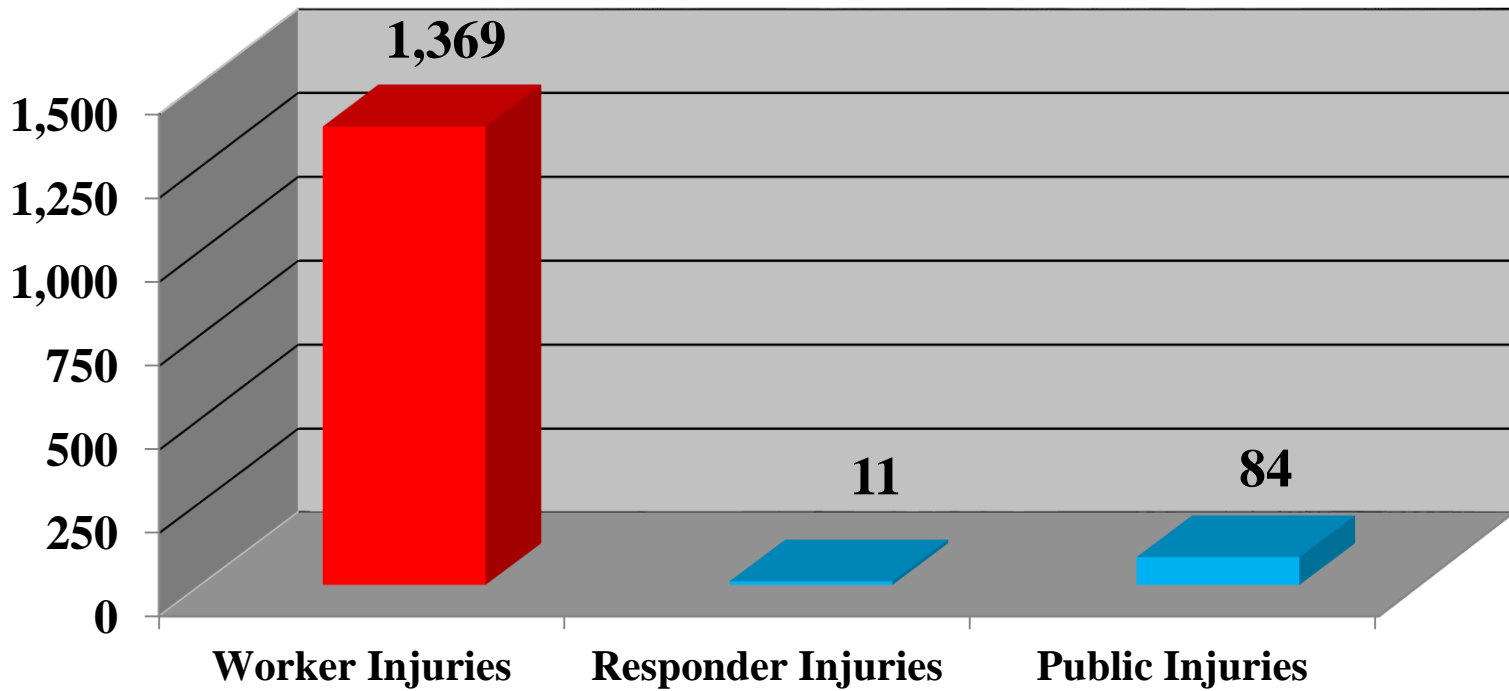


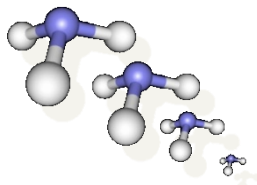
1,253 Total Releases



# Ammonia

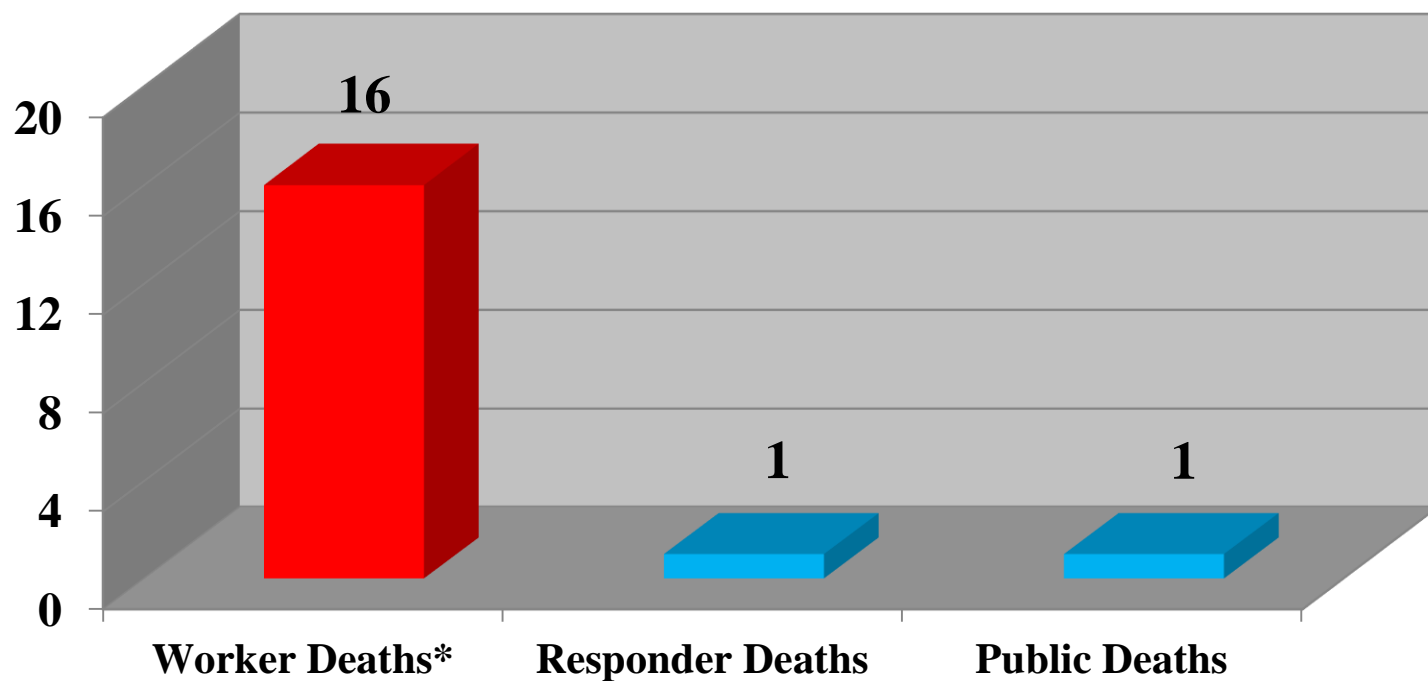
## RMP Incident Injuries (1994-2013)



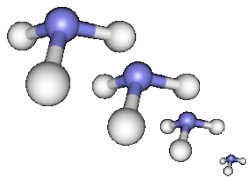


# Ammonia

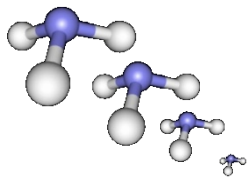
## RMP Incident Fatalities (1994 – 2013)



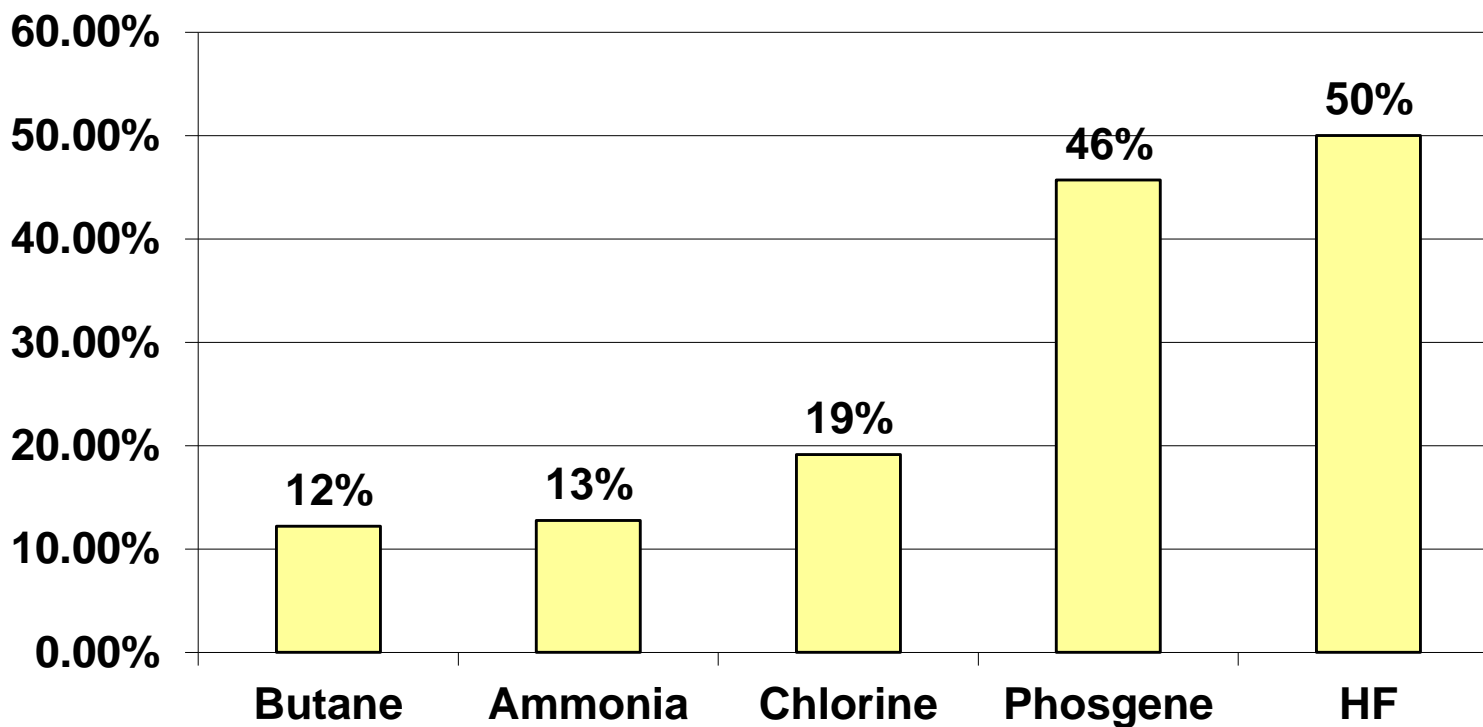


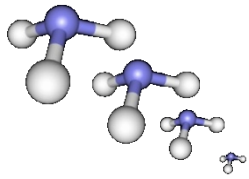


# SOME COMPARISONS

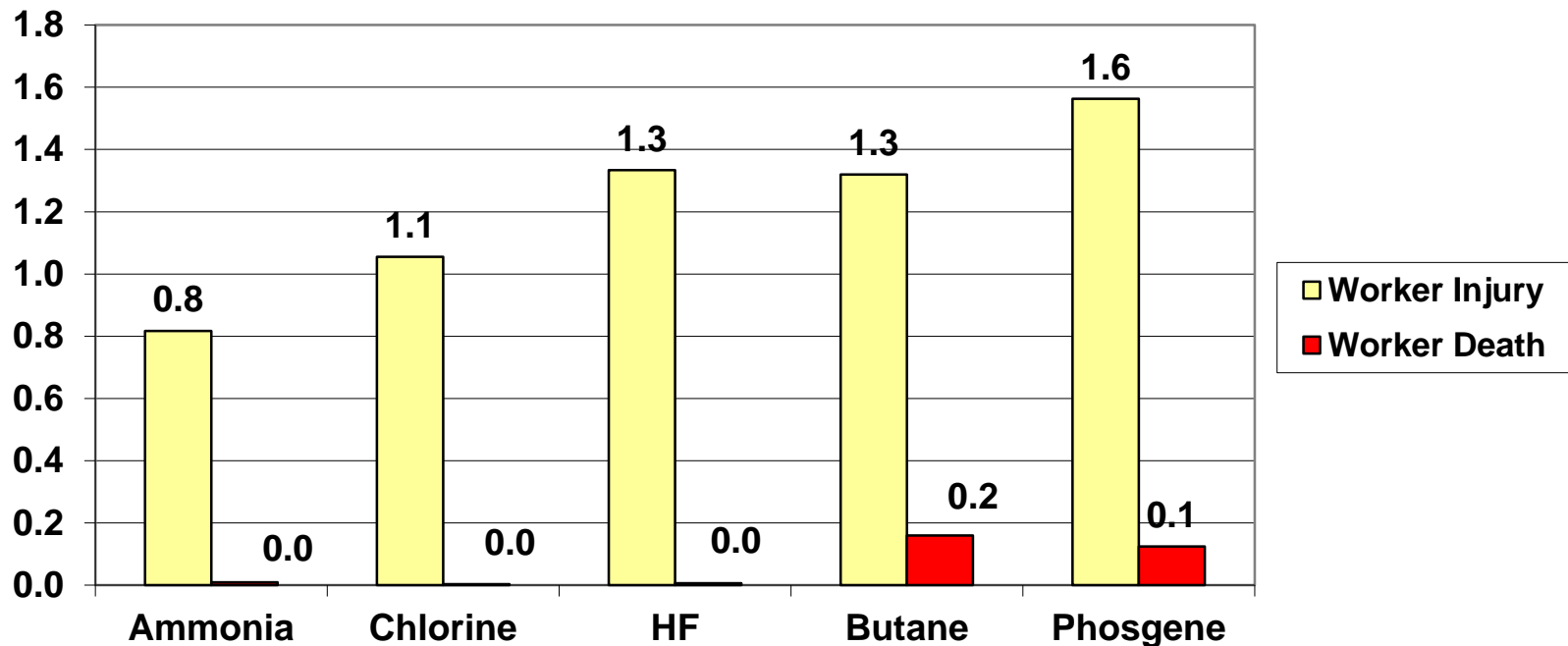


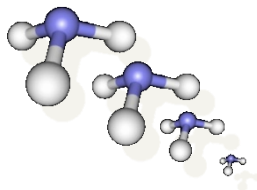
# RMP Incidents / Facility By Chemical





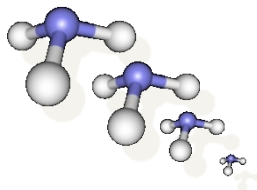
# Average Injury/Fatality per Incident by Chemical





# Conclusions

- Ammonia is one of the highest volume chemicals produced/used – in the top 5.
- Ammonia has been safely used for >150 years in tens of thousands of locations.
- Large, catastrophic accidental releases are rare and those that happened were preventable with safe design and good practices.
- The greatest risk of harm is to employees and workers exposed to high concentrations and closest to a release.



## Conclusions – cont.

- Large volume storage and handling at low pressure is safer and less risk than pressurized storage.
- Assessments of risk and consequences must be credible, based on realistic assumptions and without overly conservative criteria.
- Ammonia is a chemical that deserves respect, but it should not be feared.